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<https://research.umkc.edu/atmoschem/>



**From Farm to City:  
Multi-Scale Atmospheric Chemistry  
Relevant to Midwestern Air Quality**

Amy Christiansen, PhD

Toan Vo, Shreeram Ojha (graduate students), Dr. Luke Monroe (postdoc)

April 15, 2026

# ~7 million deaths linked to air pollution globally 4<sup>th</sup>-ranking mortality risk factor worldwide

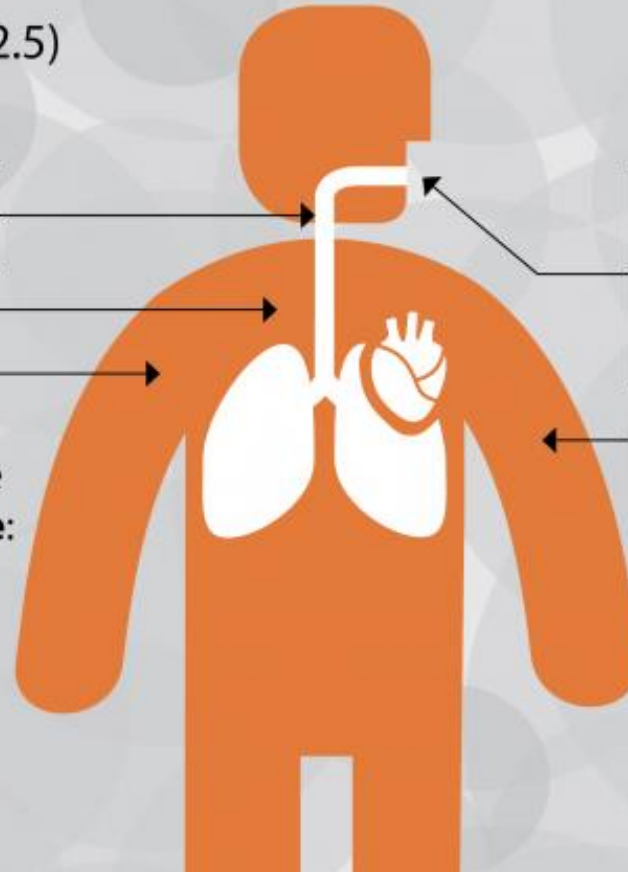


## Fine particles (PM 2.5) pollution can cause:

- Shortness of breath
- Wheezing, coughing
- Chest pain
- Fatigue

## Fine particles can make these conditions **worse**:

- Cardiovascular and heart disease
- Asthma and COPD



## Ground-level ozone pollution can cause:

- Difficulty breathing deeply
- Shortness of breath
- Sore throat
- Wheezing, coughing
- Fatigue

## Ozone can make these conditions **worse**:

- Asthma and COPD
- Emphysema

<https://www.pca.state.mn.us/air/why-you-should-care-air-quality-and-health>

<https://time.com/4167351/beijing-air-quality-pollution/>

<https://www.bbc.com/news/world-asia-india-58405479>

EPA: <https://www.epa.gov/pm-pollution/particulate-matter-pm-basics>

# Clean Air Act

Introduced in 1970

- Environmental Protection Agency (EPA) formed in 1970 to implement requirements from this Act

Major revisions in 1977 and 1990

Requires EPA to establish **national ambient air quality standards (NAAQS)** based on latest science

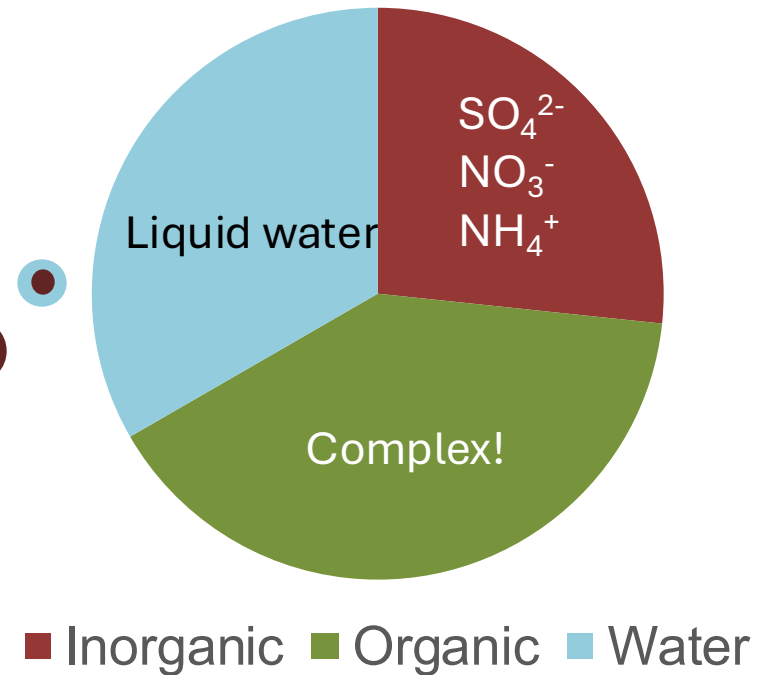
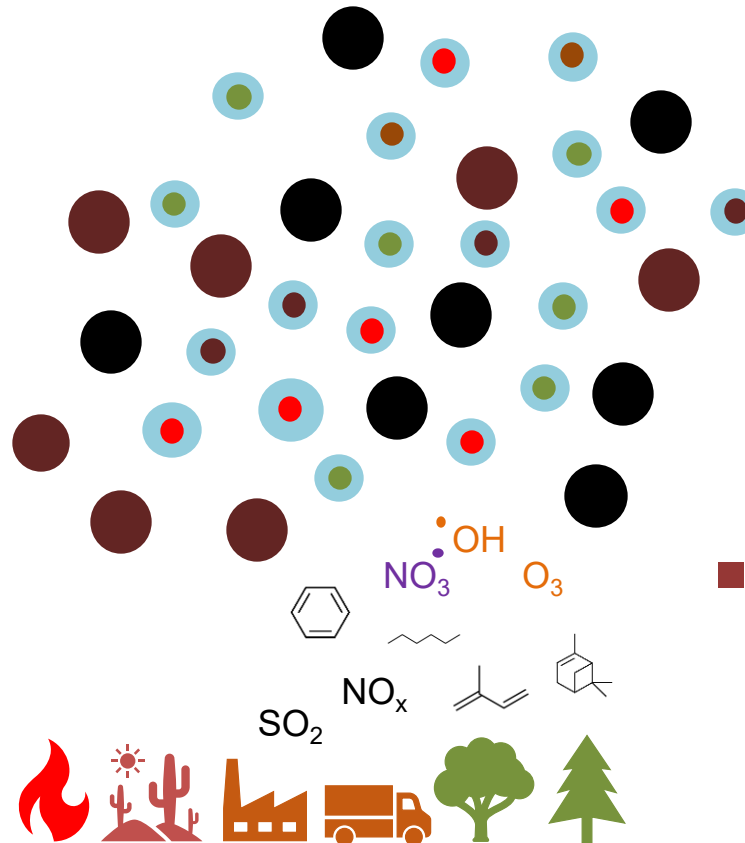
States must adopt enforceable plans to achieve and maintain air quality meeting these standards, and they must control emissions that drift across state lines

Also regulates vehicles, power plants, factories

Pollutant [links to historical tables of NAAQS reviews]		Primary/ Secondary	Averaging Time	Level	Form
<a href="#">Carbon Monoxide (CO)</a>		primary	8 hours	9 ppm	Not to be exceeded more than once per year
			1 hour	35 ppm	
<a href="#">Lead (Pb)</a>		primary and secondary	Rolling 3 month average	0.15 µg/m <sup>3</sup> <sup>(1)</sup>	maximum arithmetic mean of 3 consecutive monthly means in a 3-year period
<a href="#">Nitrogen Dioxide (NO<sub>2</sub>)</a>		primary	1 hour	100 ppb	Annual 98th percentile of 1-hour daily maximum concentrations, averaged over 3 years
		primary and secondary	1 year	53 ppb <sup>(2)</sup>	Annual Mean
<a href="#">Ozone (O<sub>3</sub>)</a>		primary and secondary	8 hours	0.070 ppm <sup>(3)</sup>	Annual fourth-highest daily maximum 8-hour concentration, averaged over 3 years
<a href="#">Particle Pollution (PM)</a>	PM <sub>2.5</sub>	primary	1 year	9.0 µg/m <sup>3</sup>	annual mean, averaged over 3 years
		secondary	1 year	15.0 µg/m <sup>3</sup>	annual mean, averaged over 3 years
		primary and secondary	24 hours	35 µg/m <sup>3</sup>	98th percentile, averaged over 3 years
	PM <sub>10</sub>	primary and secondary	24 hours	150 µg/m <sup>3</sup>	Not to be exceeded more than once per year on average over 3 years
<a href="#">Sulfur Dioxide (SO<sub>2</sub>)</a>		primary	1 hour	75 ppb <sup>(4)</sup>	Annual 99th percentile of 1-hour daily maximum concentrations, averaged over 3 years
		secondary	1 year	10 ppb	annual mean, averaged over 3 years

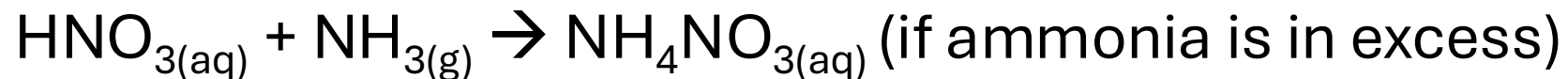
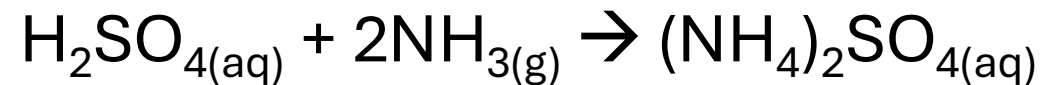
# What is particulate matter (PM)?

- Small solid or liquid particles that are suspended in air
- Primary and secondary sources
  - Primary: Dust, sea spray, soot (**larger, PM<sub>10</sub>**)
  - Secondary: Oxidation of gas-phase precursor species (**smaller, PM<sub>2.5</sub>**)
- Composed of complex chemical matrix



# New Particle Formation

- Spontaneous condensation of vapors (homogeneous nucleation) unlikely due to Kelvin effect (vapor pressure increases drastically on a curved surface)
- Nucleation involves condensation onto pre-existing clusters of molecules (heterogeneous nucleation)
- Acid-base reactions dominate



- Condensation of other vapors onto existing particle
  - Organics dominate

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**ambient air quality**  
 based on latest

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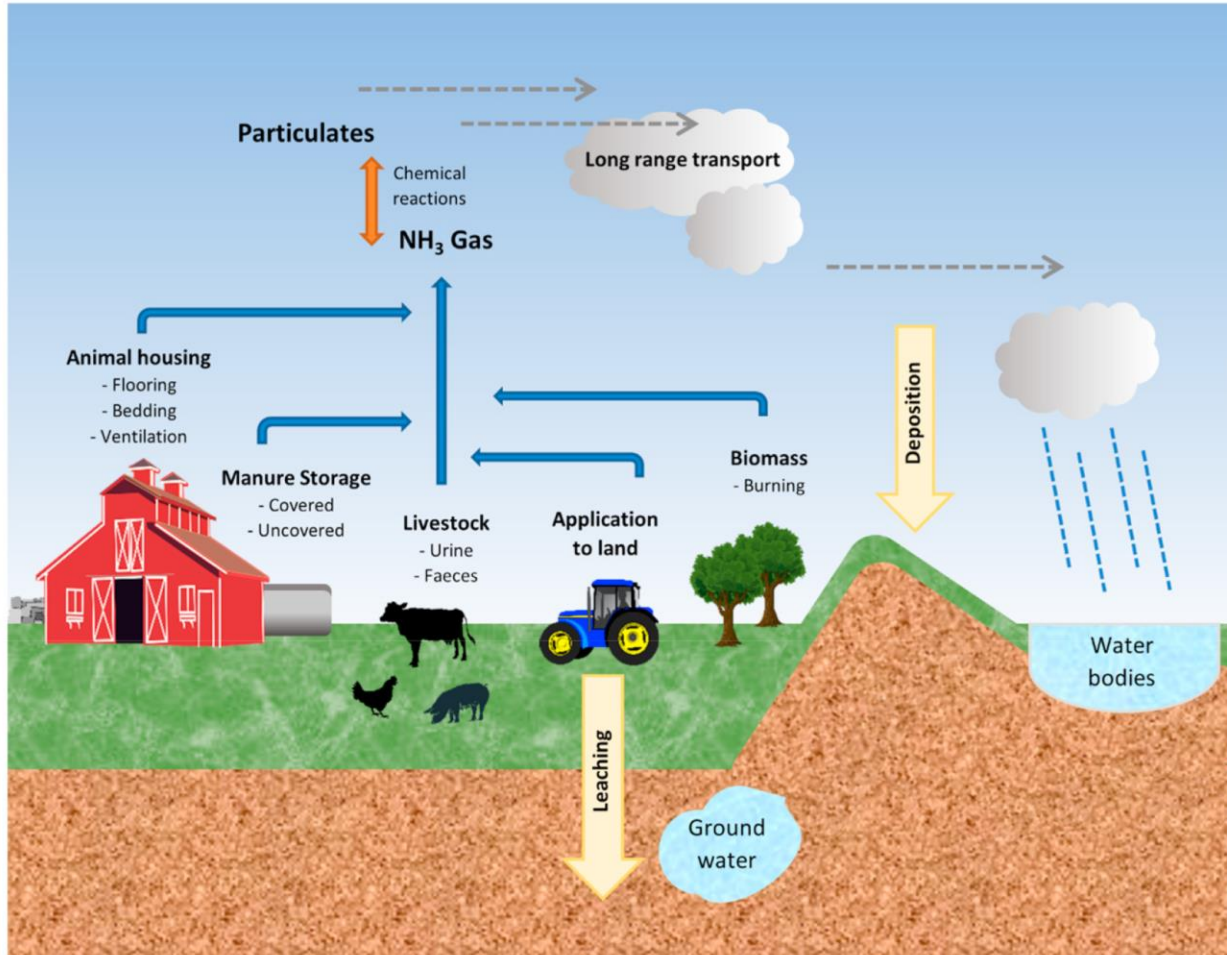
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## Impact of Ammonia??

fourth-highest daily maximum 8-hour concentration, averaged over 3 years

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<a href="#">Sulfur Dioxide (SO<sub>2</sub>)</a>	primary	1 hour	75 ppb <sup>(4)</sup>	Annual 99th percentile of 1-hour daily maximum concentrations, averaged over 3 years	
	secondary	1 year	10 ppb	annual mean, averaged over 3 years	

# Agricultural Emissions



- Globally, agricultural ammonia (NH<sub>3</sub>) emissions have increased by 78% from 1980 to 2018
- ~50% over Midwestern United States since 2001
- The Midwestern United States (MWUS) is one of the world's most agriculturally intense regions and accounts for ~40% of the country's agricultural NH<sub>3</sub> emissions
- NH<sub>3</sub> serves as a key precursor gas in the formation of fine particulate matter (PM<sub>2.5</sub>)
- Agricultural NH<sub>3</sub> emissions are not regulated in the US through the Clean Air Act, and the impact of agricultural NH<sub>3</sub> on downwind air quality is not well quantified
- **Goal: Determine the impact of increasing agricultural NH<sub>3</sub> on the formation of PM<sub>2.5</sub> throughout the MWUS from 2007 to 2019**

# Outline

**How have agricultural emissions impacted Midwestern air quality?**

**1) Agricultural  $\text{NH}_3$  increases  $\text{PM}_{2.5}$  burden in the Midwest disproportionately compared to the contiguous US (CONUS)**

How have the chemical driving factors behind particulate nitrate formation changed over time? Can this be used to find the optimal method for controlling  $\text{PM}_{2.5}$  burden?

**Emissions**  
**Meteorology**

**Concentrations**

Simulations	Standard	AgrOFF	NH32007	Met2007
Emissions	All emissions included	Excluded agricultural NH <sub>3</sub> emissions	Held agricultural NH <sub>3</sub> emissions constant at 2007 values	All emissions included
Meteorology	Standard meteorology	Standard meteorology	Standard meteorology	Held meteorology constant at 2007 values

**Transport**  
**Chemistry**

$$\frac{\partial n_i}{\partial t} = -\nabla \cdot (n_i \mathbf{U}) + P_i - L_i$$

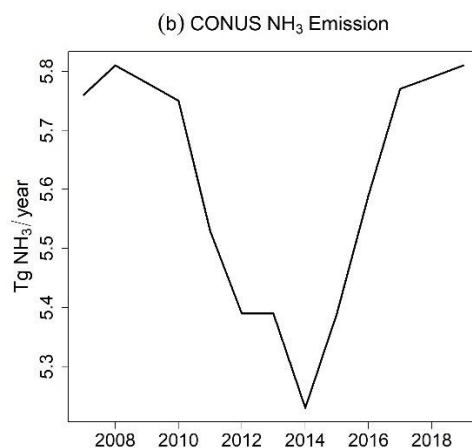
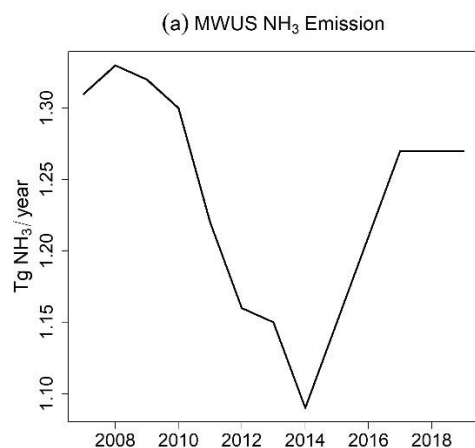
local trend in concentration

transport (flux divergence)

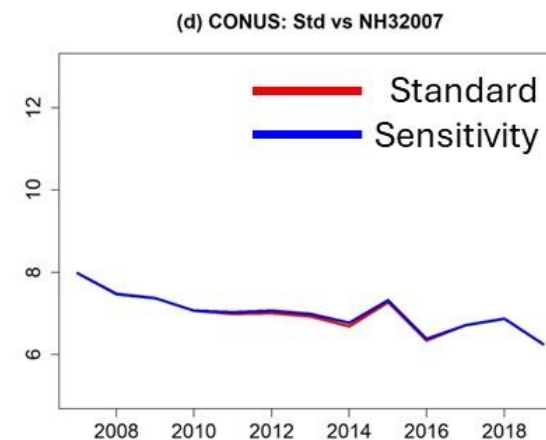
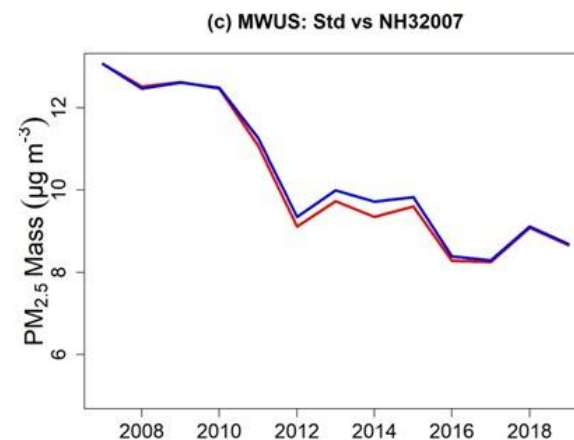
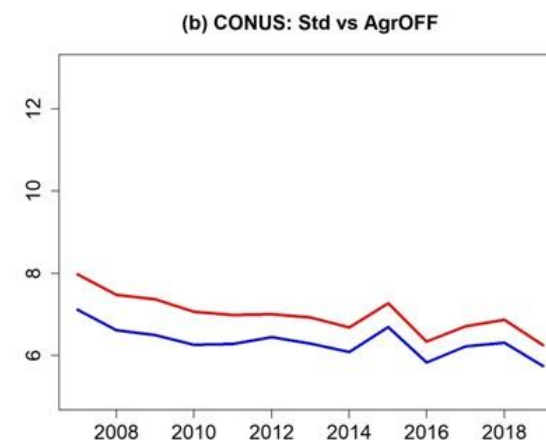
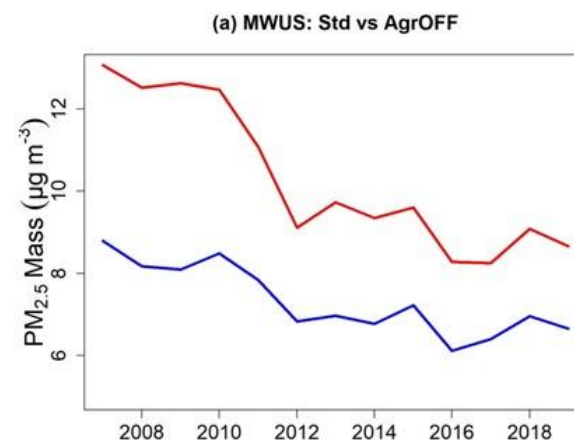
emissions, deposition, chemical and aerosol processes

- When agricultural emissions turned off (AgrOFF)
  - CONUS PM<sub>2.5</sub> decreases by 9% (sensitivity decreases (Std – AgrOFF))
  - MWUS PM<sub>2.5</sub> decreases by 29%
  - Agricultural NH<sub>3</sub> contributes 10% to CONUS PM<sub>2.5</sub>, but 40% to MWUS PM<sub>2.5</sub>

- When NH<sub>3</sub> held constant at 2007 values (NH32007):
  - Sensitivity of PM<sub>2.5</sub> to NH<sub>3</sub> trend differs
  - MWUS peak to trough: 3.7% decrease
  - CONUS peak to trough: 0.5% decrease
  - Reducing NH<sub>3</sub> to 2014 levels results in a savings of 0.4 μg m<sup>-3</sup> in the MWUS, but only 0.1 μg m<sup>-3</sup> over the CONUS



## Agricultural NH<sub>3</sub> Leads to Higher PM<sub>2.5</sub> Burden in MWUS than over CONUS



# Summary

- Agricultural  $\text{NH}_3$  hinders air quality improvement efforts in the MWUS
- $\text{PM}_{2.5}$  is disproportionately higher in the MWUS compared to the CONUS, and observations and models suggest that agricultural  $\text{NH}_3$  is behind this discrepancy
  - Largest impacts occur in winter
- GEOS-Chem simulations suggest that  $\text{PM}_{2.5}$  burden could be lowered by  $0.4 \mu\text{g m}^{-3}$  (3.5%) in the MWUS through reductions in  $\text{NH}_3$  emissions to 2014 levels (20% decrease in emissions)
- Continued increases in agricultural  $\text{NH}_3$  are a public health concern, and  $\text{NH}_3$  should be regulated as a criteria pollutant

# Outline

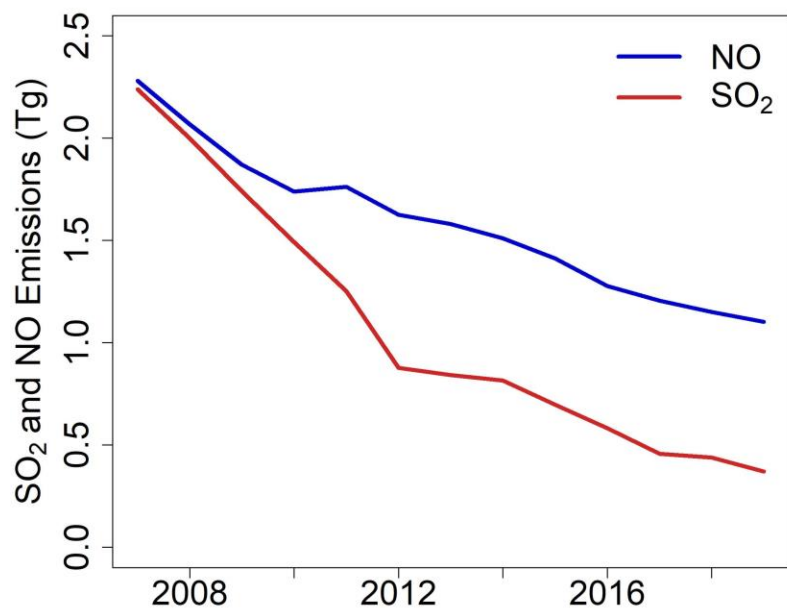
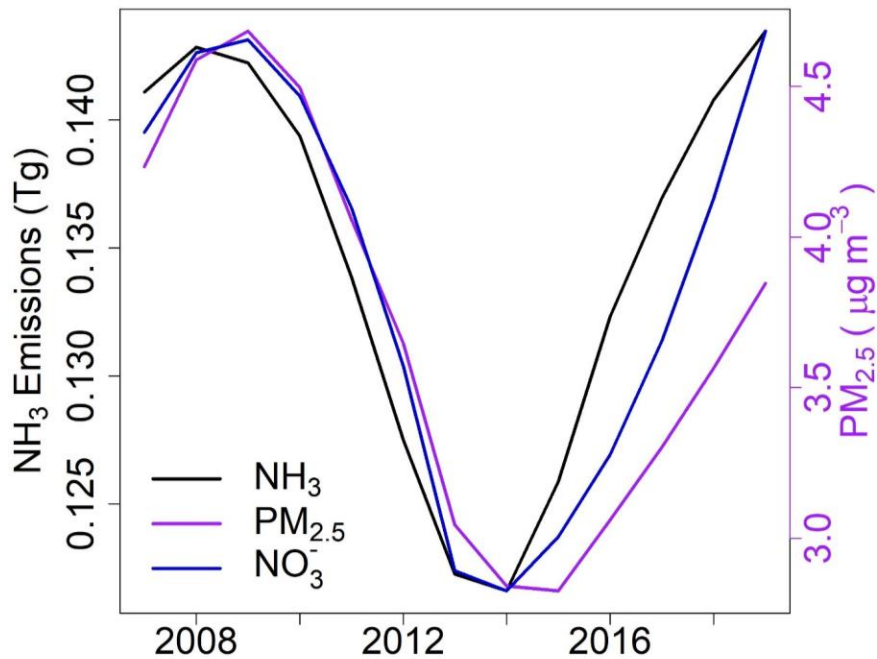
How have agricultural emissions impacted Midwestern air quality?

- 1) Agricultural  $\text{NH}_3$  increases  $\text{PM}_{2.5}$  burden in the Midwest disproportionately compared to the CONUS

**How have the chemical driving factors behind particulate nitrate (PN) formation changed over time? Can this be used to find the optimal method for controlling  $\text{PM}_{2.5}$  burden?**

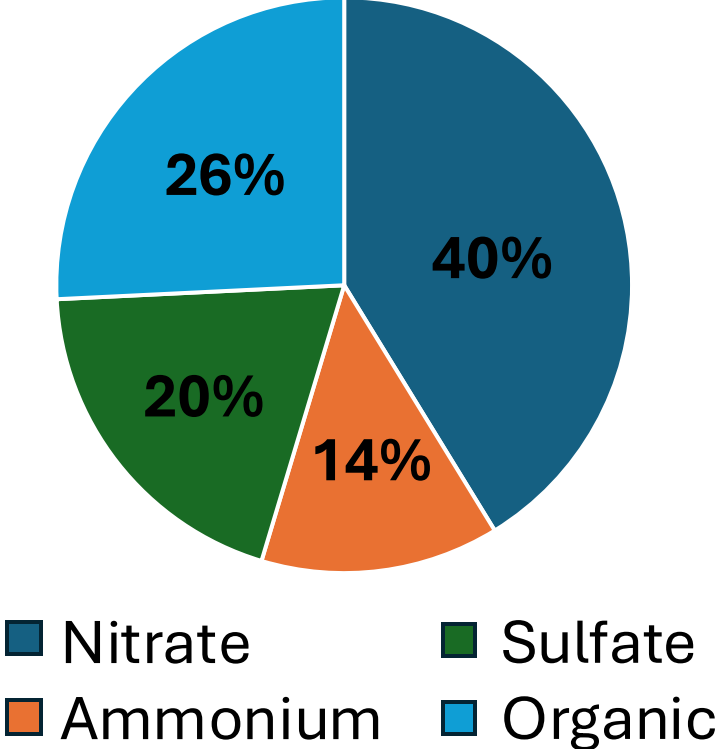
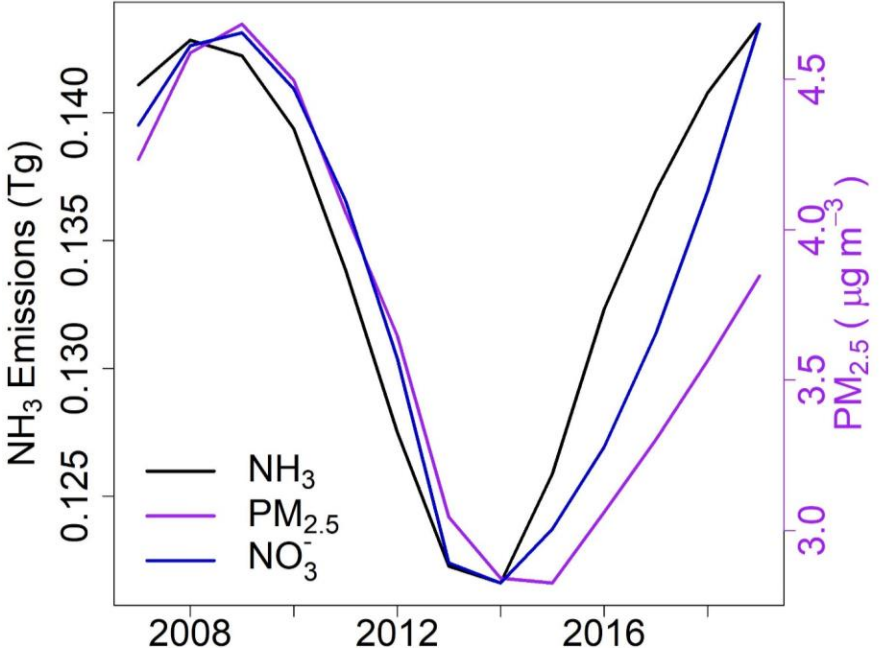
- 1) Wintertime particulate nitrate formation has become more sensitive to nitrogen oxides over time.
- 2) Thus, controlling nitrogen oxide emissions will best control  $\text{PM}_{2.5}$  burden.

## Wintertime $\text{NH}_3$ Emissions Increases Coincide with Increases in $\text{PM}_{2.5}$ and Nitrate



- Prior to 2014,  $\text{PM}_{2.5}$  and  $\text{NO}_3^-$  from IMPROVE network show decrease in concentrations
- After 2014,  $\text{PM}_{2.5}$  and  $\text{NO}_3^-$  trends begin to reverse
  - Trend only exists in winter, when semi-volatile  $\text{NO}_3^-$  is most thermodynamically stable in the particle
- Reversal of  $\text{NO}_3^-$  and  $\text{PM}_{2.5}$  trend may be caused by increase in  $\text{NH}_3$  emissions
  - Not from  $\text{NO}$  or  $\text{SO}_2$ !

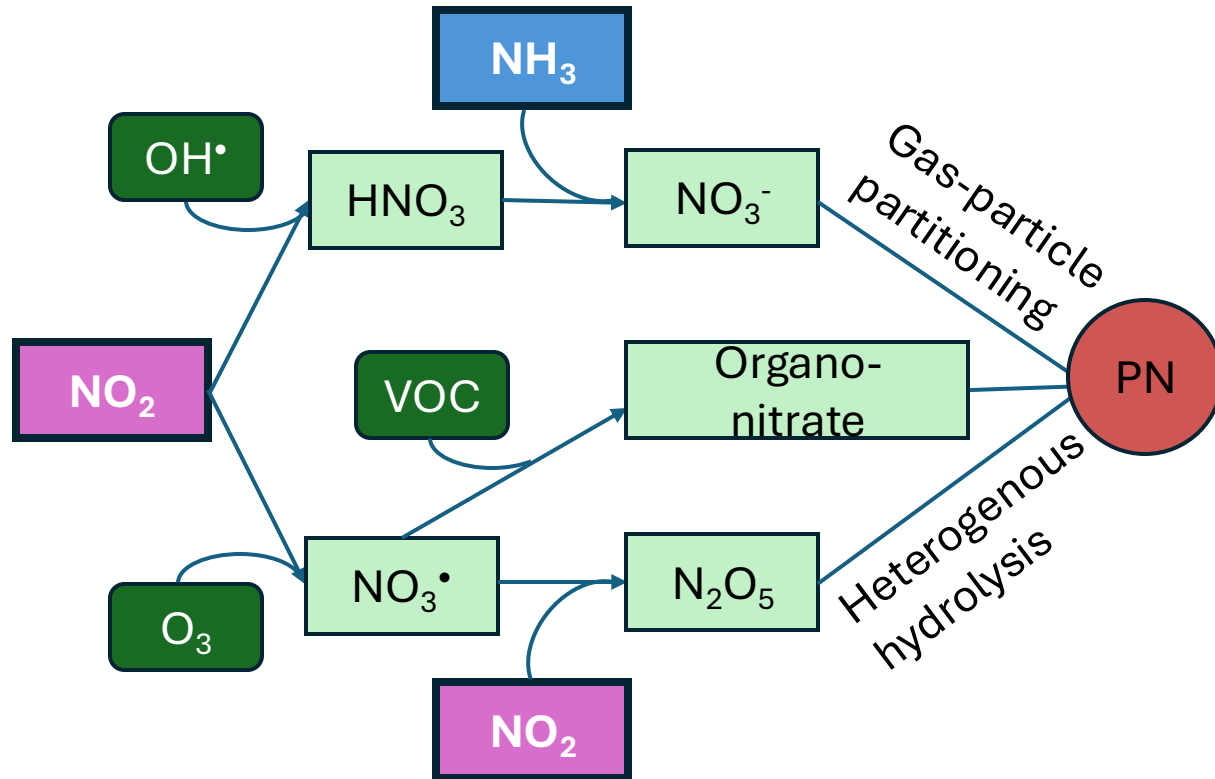
# Controlling Wintertime PN is Critical to Reducing PM<sub>2.5</sub> Burden over the MWUS



PN drives many pollution events and controls water uptake processes

Impacting human health, weather and climate

# PN Formation Sensitivity



PN formation is sensitive to whichever gas is the limiting reagent ( $\text{NO}_x$ ,  $\text{NH}_3$ , or  $\text{VOC}$ )

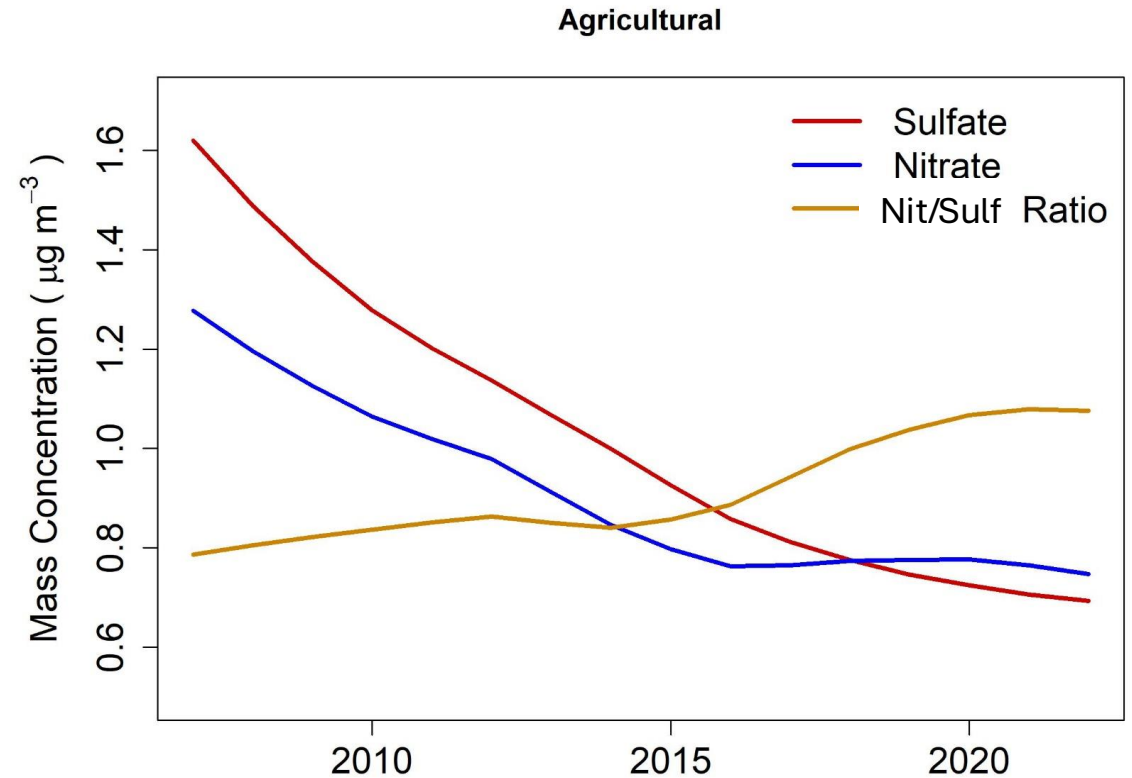
Previous methods in diagnosing PN sensitivity are highly uncertain and computationally intensive

Satellite observations can overcome the limitations

# Goal

Quantify the changing sensitivity of wintertime PN to  $\text{NH}_3$ ,  $\text{NO}_x$ , and VOCs over the MWUS from 2007 to 2023 using satellite observations, ground monitoring data, and model simulations.

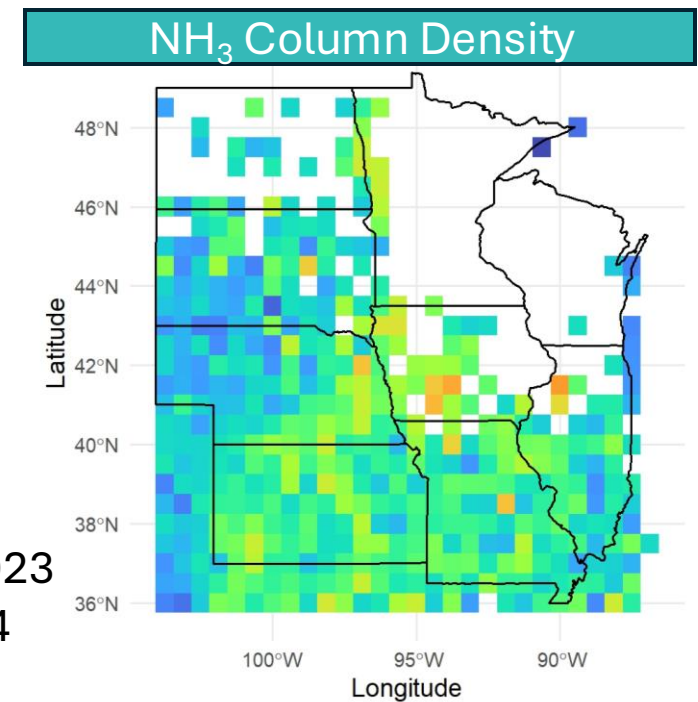
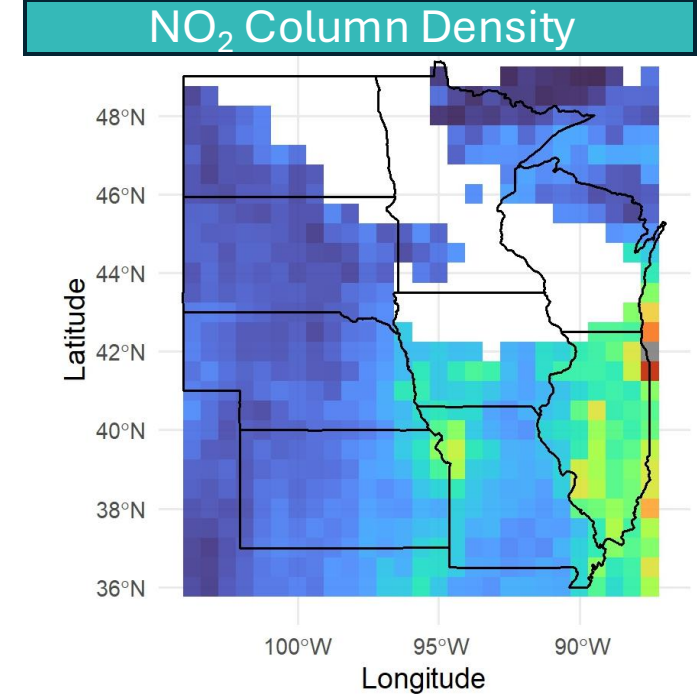
Understanding the sensitivity of PN formation in winter over the MWUS provides insights to the most effective  $\text{PM}_{2.5}$  mitigation strategy.



# Diagnosing Wintertime PN Sensitivity

- Satellite retrievals for  $\text{NO}_2$  and  $\text{NH}_3$  column densities (2007-2023)
  - Ozone Monitoring Instrument (OMI) and Infrared Atmospheric Sounding Interferometer (IASI)
- Averaged to a  $0.5^\circ \times 0.625^\circ$  resolution to match GEOS-Chem simulations
- Calculated satellite tropospheric column  $\text{NH}_3/\text{NO}_2$  ratios

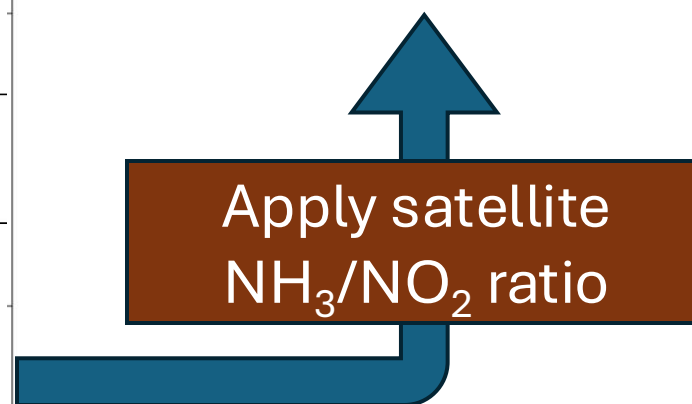
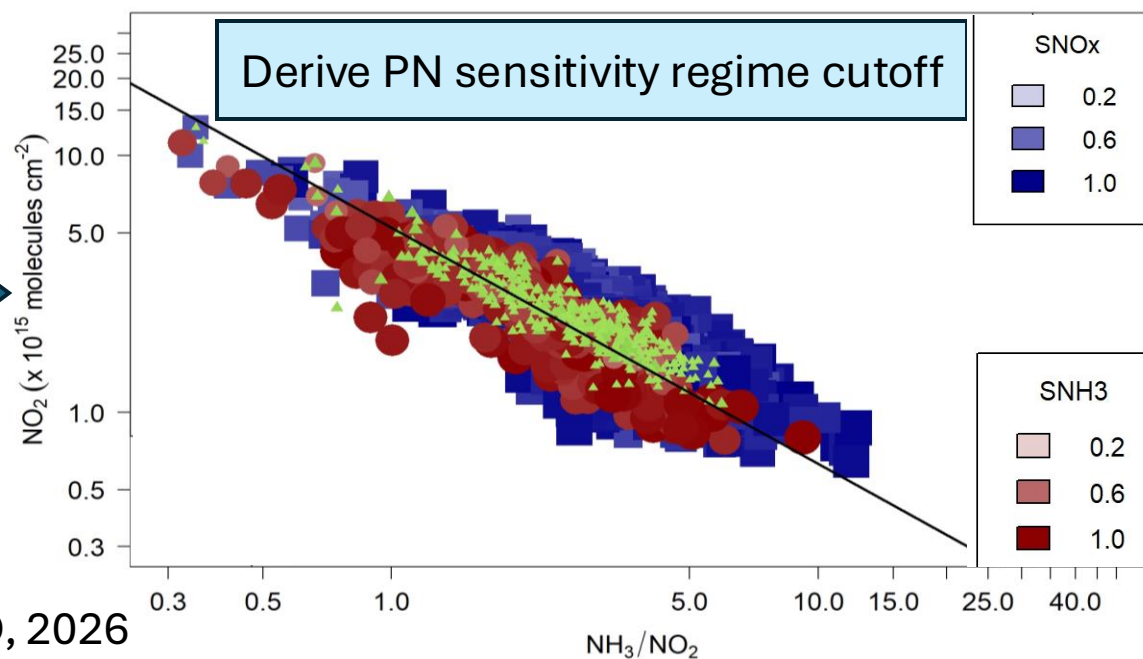
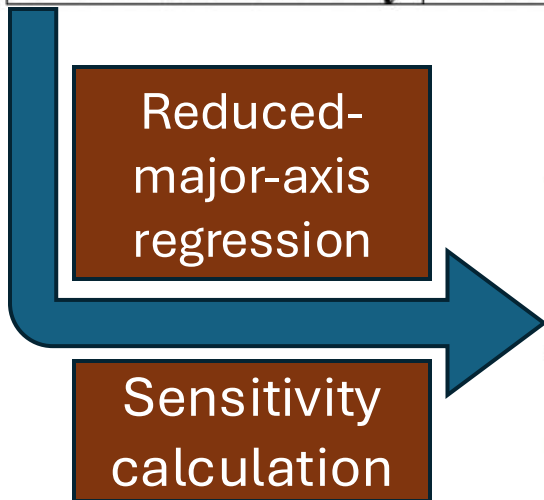
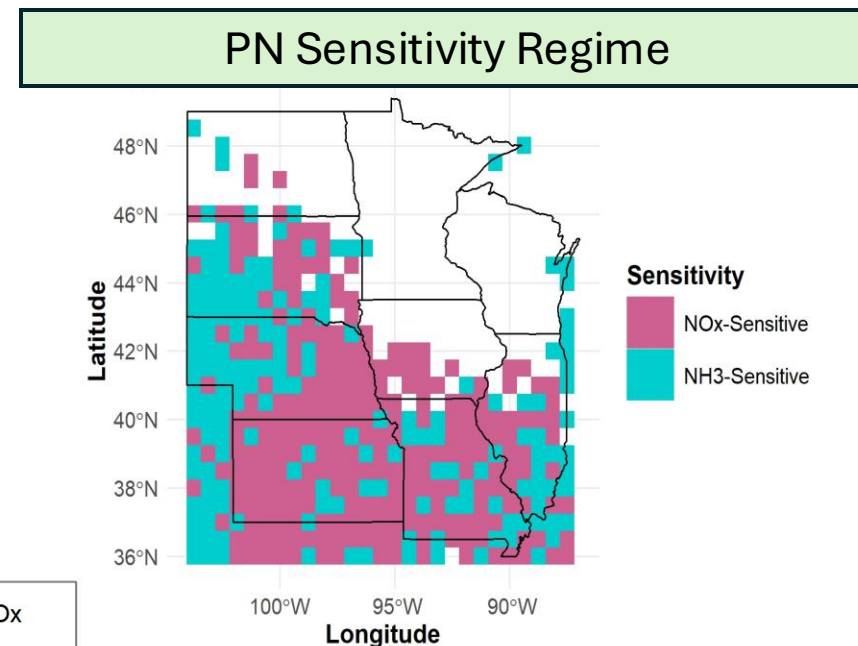
Dang et al., *Geophys. Res. Lett.*, 2023  
Dang et al., *Env. Sci. Technol.*, 2024  
Vo and Christiansen, *ACPD*, 2026



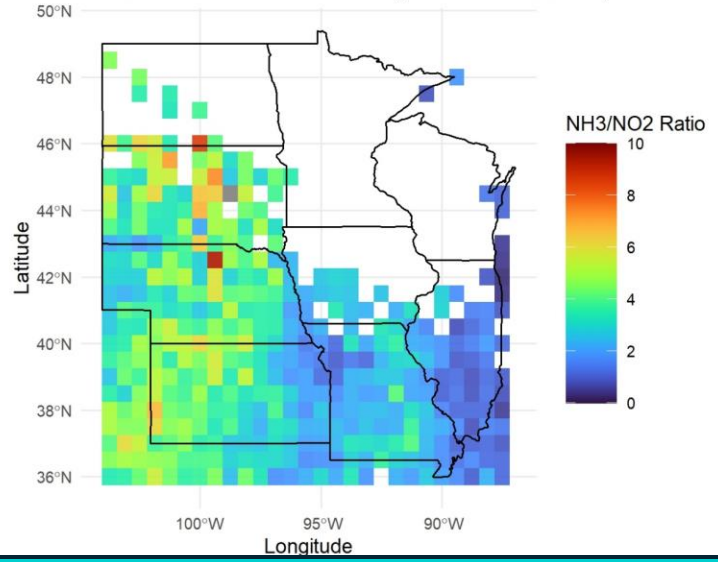
# Diagnosing Wintertime PN Sensitivity

GEOS-Chem sensitivity simulations			
Simulation	NO <sub>x</sub>	NH <sub>3</sub>	VOCs
Base	Normal	Normal	Normal
NO <sub>x</sub> Sensitivity	-20%	Normal	Normal
NH <sub>3</sub> Sensitivity	Normal	-20%	Normal
VOC Sensitivity	Normal	Normal	-20%

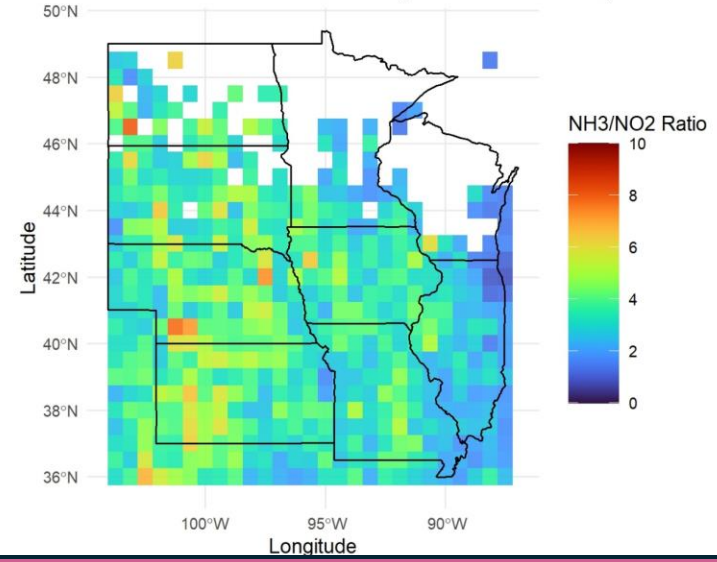
$$S_i = \frac{\Delta \log(PN)}{\Delta \log(E_i)}$$



NH3/NO2 Ratio: Winter 2007 (Corrected OMNO2)



NH3/NO2 Ratio: Winter 2023 (Corrected OMNO2)



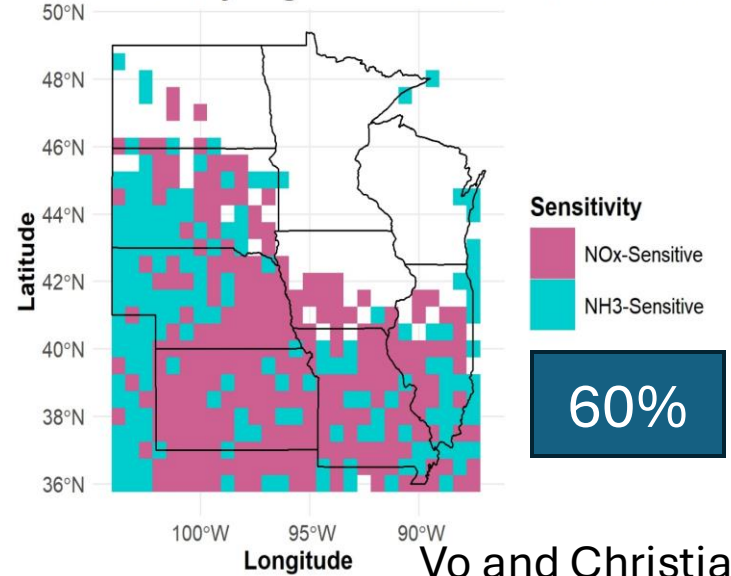
PN sensitivity over the MWUS has shifted to a predominantly NO<sub>x</sub>-sensitive regime.

*For NH<sub>3</sub> – sensitive regime:*  
 $\log\left(\frac{NH_3}{NO_2}\right) < 0.72 - 0.92 \times \log(NO_2)$

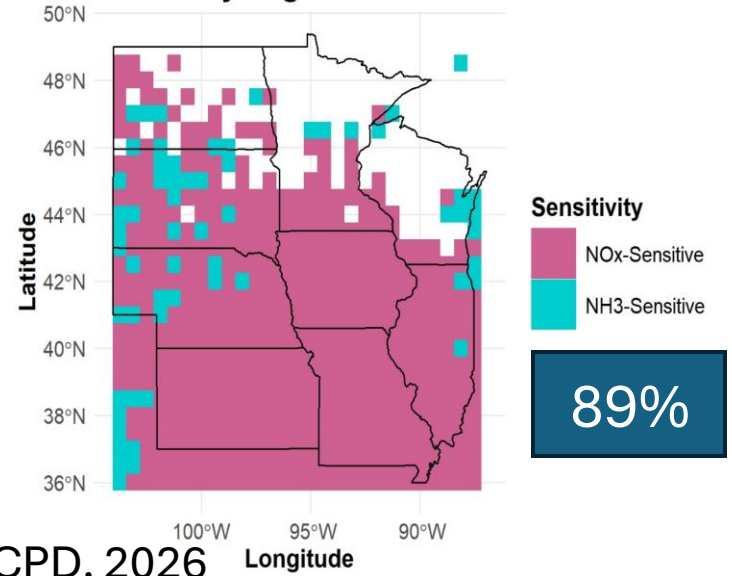
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Sensitivity Regime over MWUS - Winter 2007

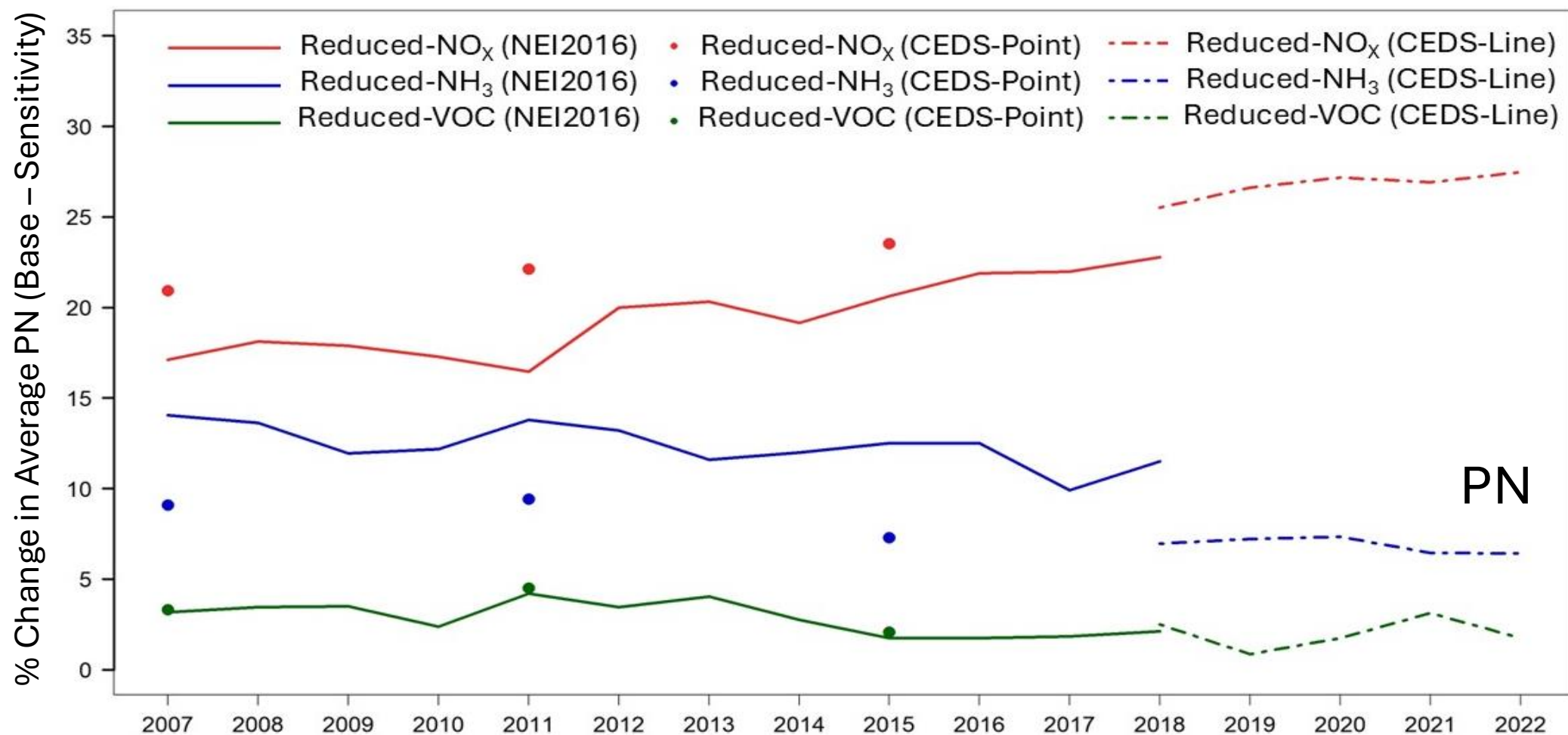


Sensitivity Regime over MWUS - Winter 2023



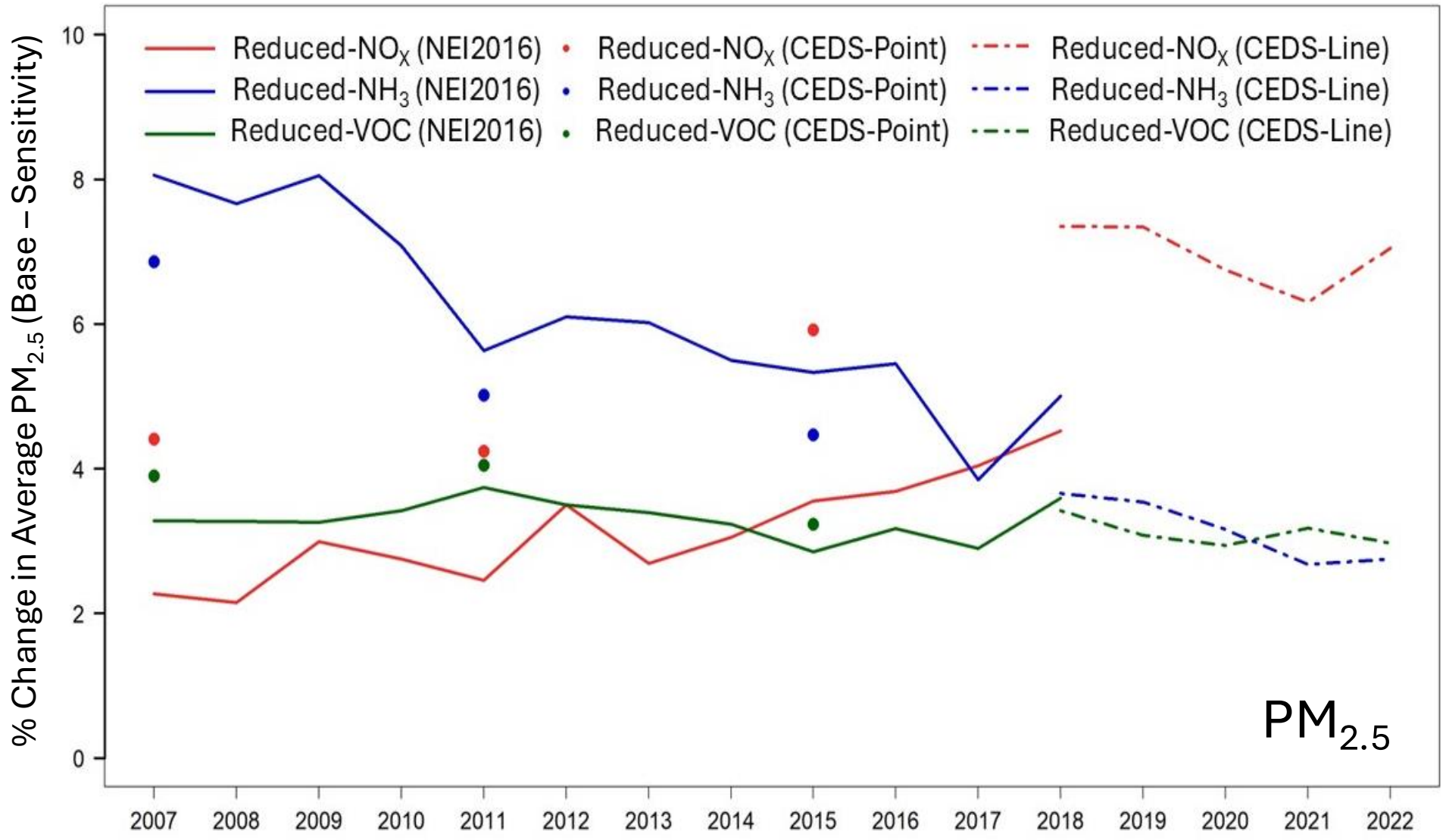
Controlling wintertime NO<sub>x</sub> emissions will effectively reduce PN over the MWUS.

# PN Sensitivity over the MWUS Has Shifted to a Predominantly $\text{NO}_x$ -sensitive Regime



PN

# Controlling Wintertime $\text{NO}_x$ Emissions Will Effectively Reduce Wintertime $\text{PM}_{2.5}$ over the MWUS



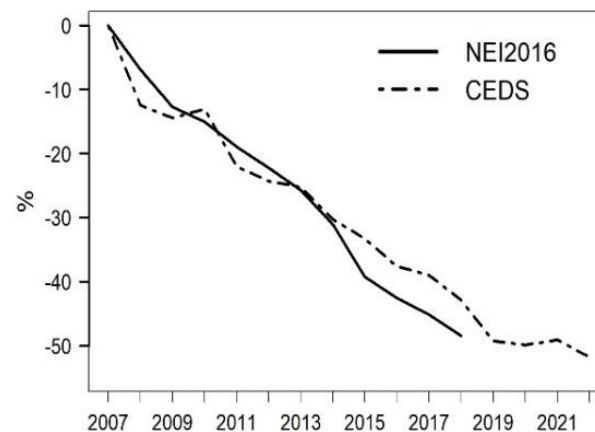
Vo and Christiansen, ACPD, 2026

$\text{PM}_{2.5}$

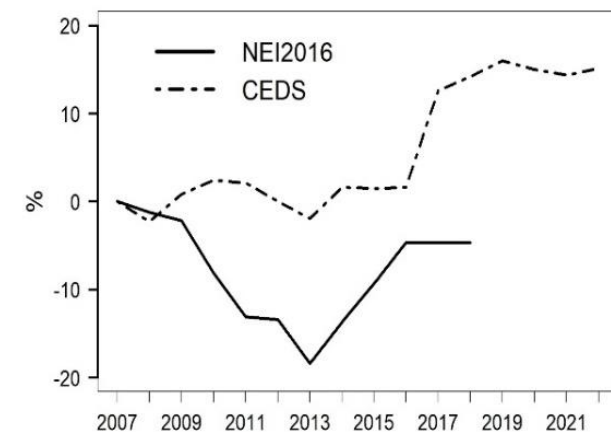
# Summary

- Wintertime PN formation is becoming more sensitive to  $\text{NO}_x$  emissions over the MWUS
  - Driven by multidecadal changes in precursor gas emissions
- Chemically, it is beneficial to target  $\text{NO}_x$  emissions to reduce wintertime PN and  $\text{PM}_{2.5}$  over MWUS
- Future work should investigate the sensitivity of PN and  $\text{PM}_{2.5}$  to precursor gas emissions in other seasons

MWUS Wintertime:  $\text{NO}_x$  Emissions (2007–2022) (NEI2016 vs. CEDS)



MWUS Wintertime:  $\text{NH}_3$  Emissions (2007–2022) (NEI2016 vs. CEDS)



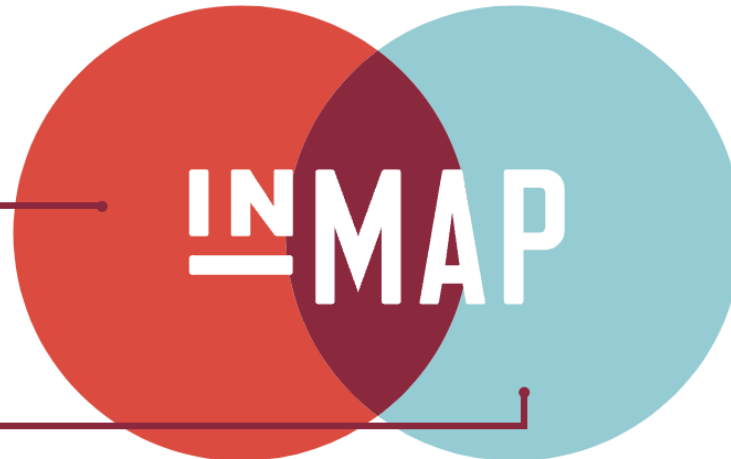
# One last thing... Kansas City-specific AQ

- InMAP
  - InMAP is a simplified air quality model
- Determines grid size dynamically:
  - High population → fine grid ( $2 \times 2 \text{ km}^2$ ): urban areas
  - Low population → coarse grid ( $12 \times 12 \text{ km}^2$ ): rural areas
- InMAP estimates annual average total  $\text{PM}_{2.5}$  concentration

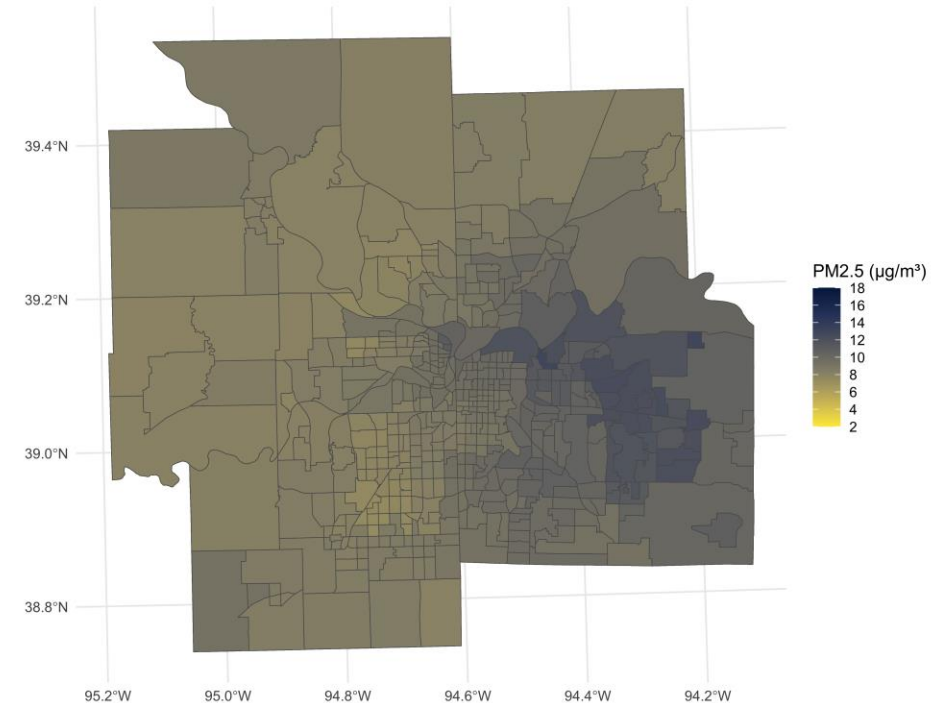
**InMAP** combines

Long range transport  
Secondary pollutant formation  
Many pollutants  
Many sources

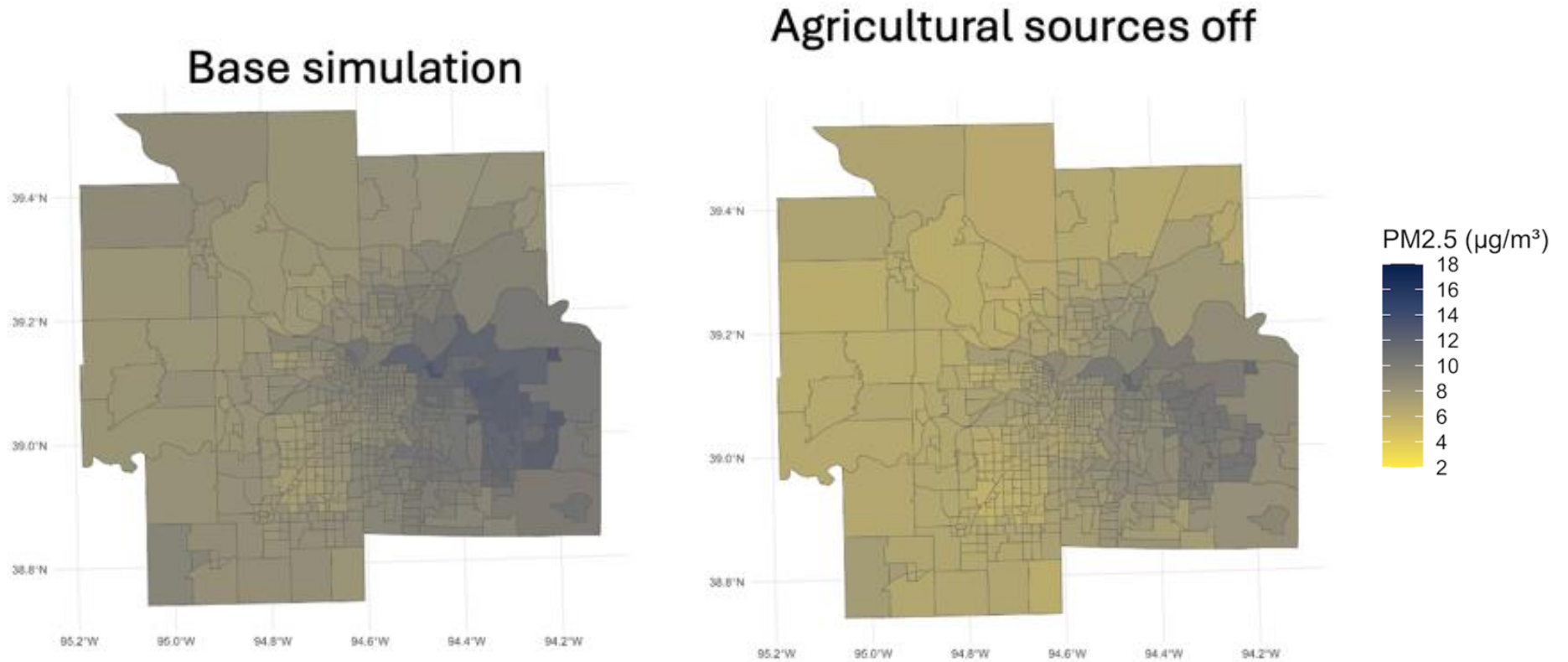
High spatial resolution  
Computational efficiency



Total  $\text{PM}_{2.5}$  Concentration: Including all sources



# Eliminating agricultural emissions results in a PM<sub>2.5</sub> reduction of 1.6 $\mu\text{g m}^{-3}$ (18%) in Kansas City





# Acknowledgements

- Christiansen group members
  - **Toan Vo**
  - **Shreeram Ojha**
  - **Dr. Luke Monroe** (now at NW Missouri State)
  - Kabin Chemjong
  - Lizbeth Gracida-Basurto
- UMKC Funding for Faculty Excellence
- US DOT
- Visit our website: <https://research.umkc.edu/atmoschem/>
- Contact me: [achristiansen@umkc.edu](mailto:achristiansen@umkc.edu)





# Harvard Six Cities Study 1993

Long-term study of six cities with varying pollution levels

Tracked rate of mortality over 14-16 years

**W**atertown, MA

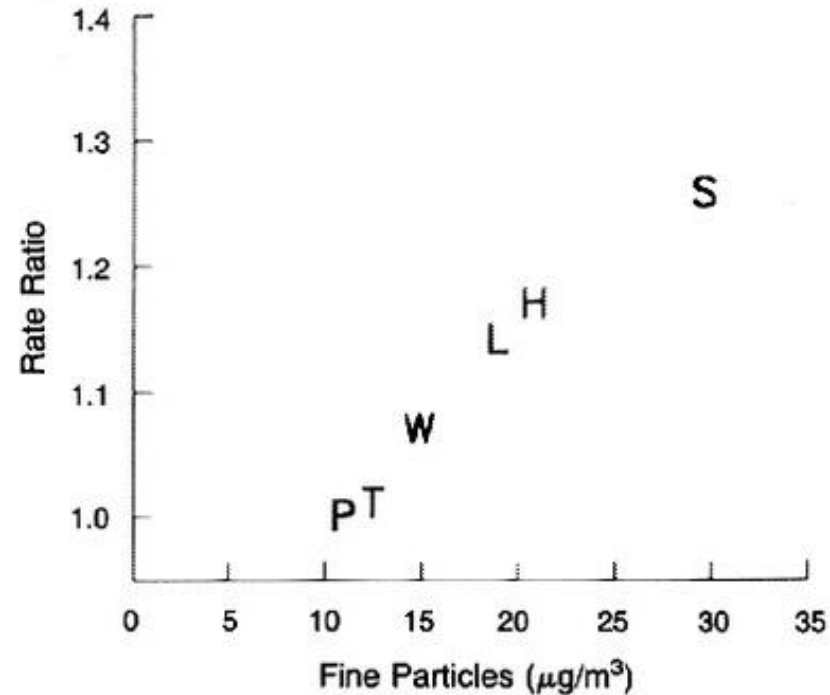
**H**arriman, TN

**S**t. Louis, MO

**S**teubenville, OH

**P**ortage, WI

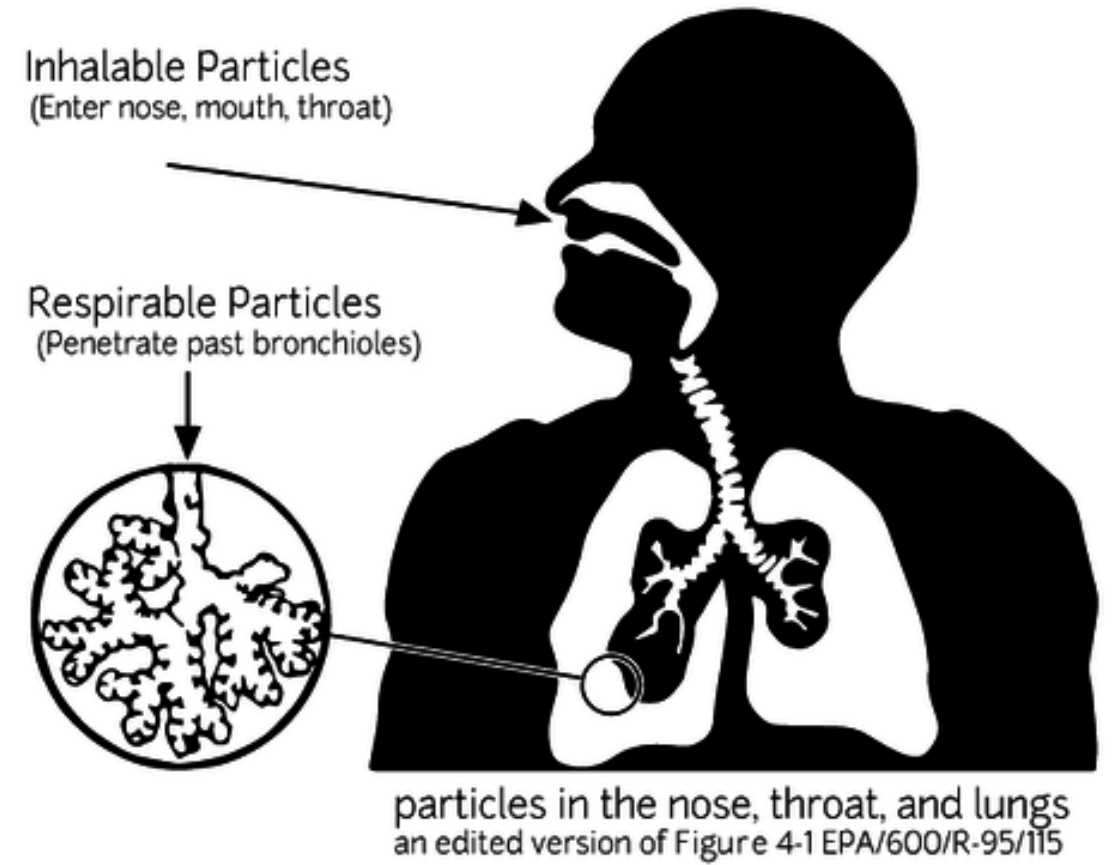
**T**opeka, KS



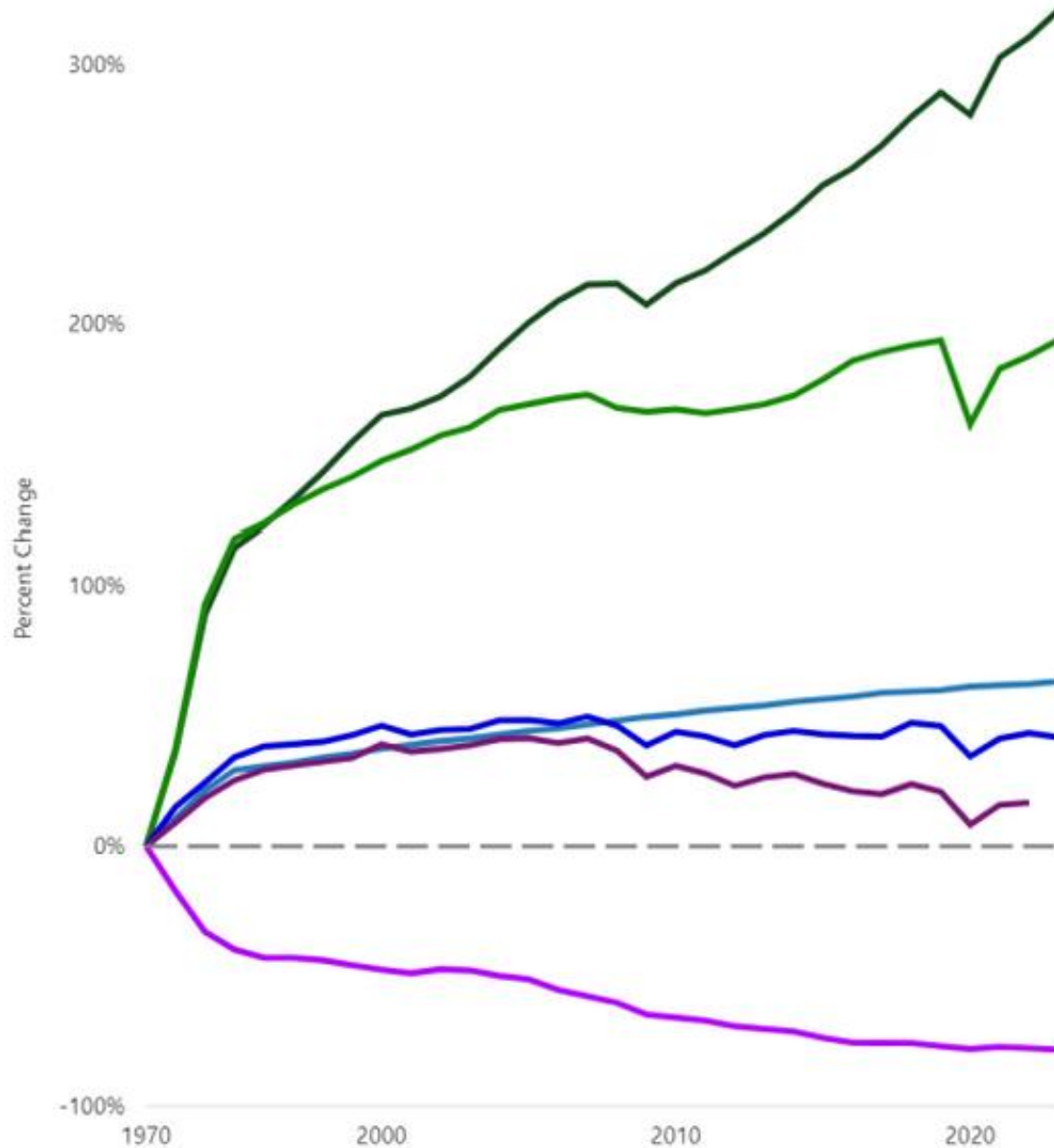
Clear association between particle mass and mortality!

# Why PM<sub>2.5</sub>?

- PM<sub>2.5</sub> can get past protective layer of lungs (cilia) and enter bloodstream
  - Oxidative stress, inflammatory response → disease

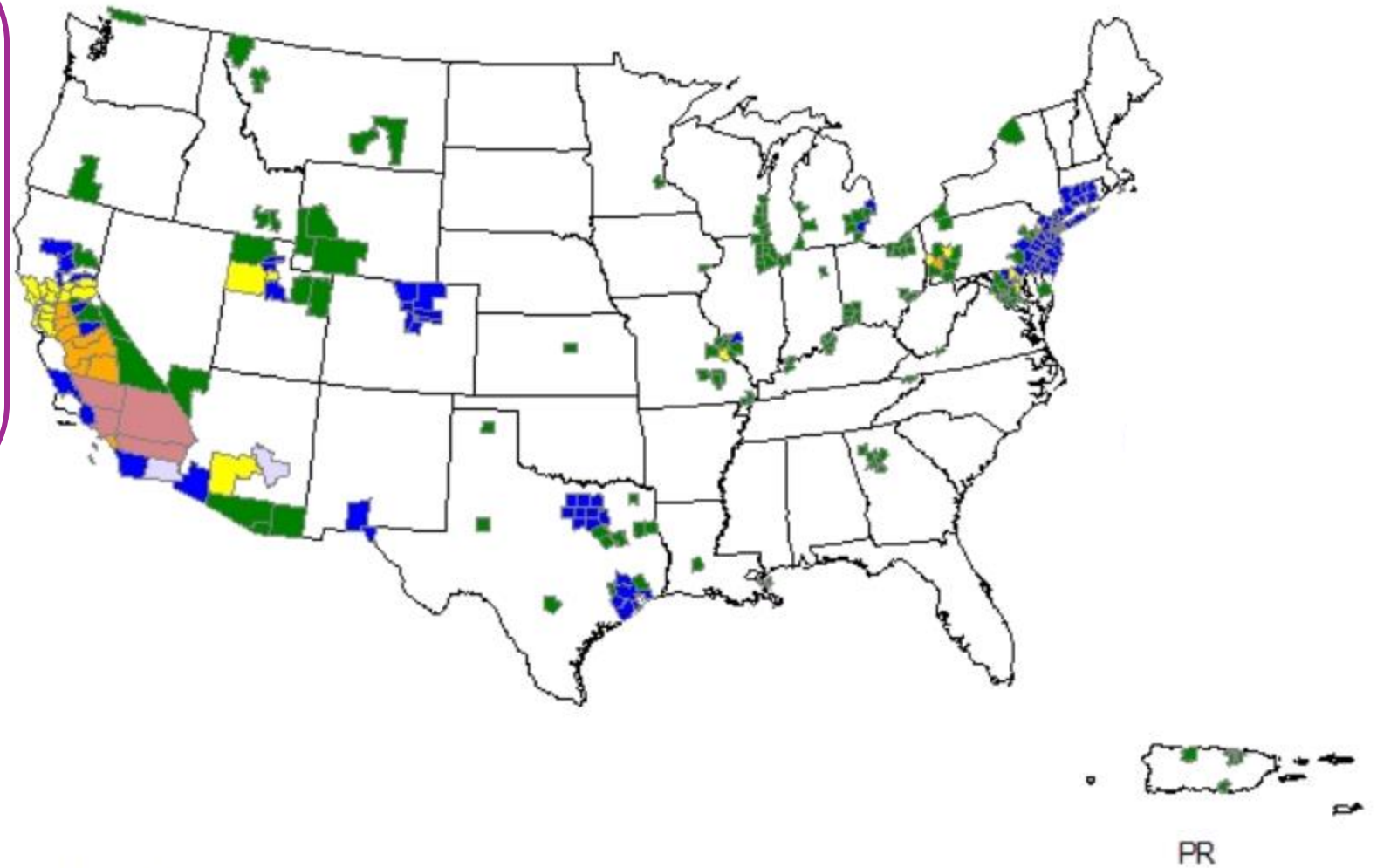


# Comparison of Growth Areas and Declining Emissions 1970-2023



Air pollution problems persist today

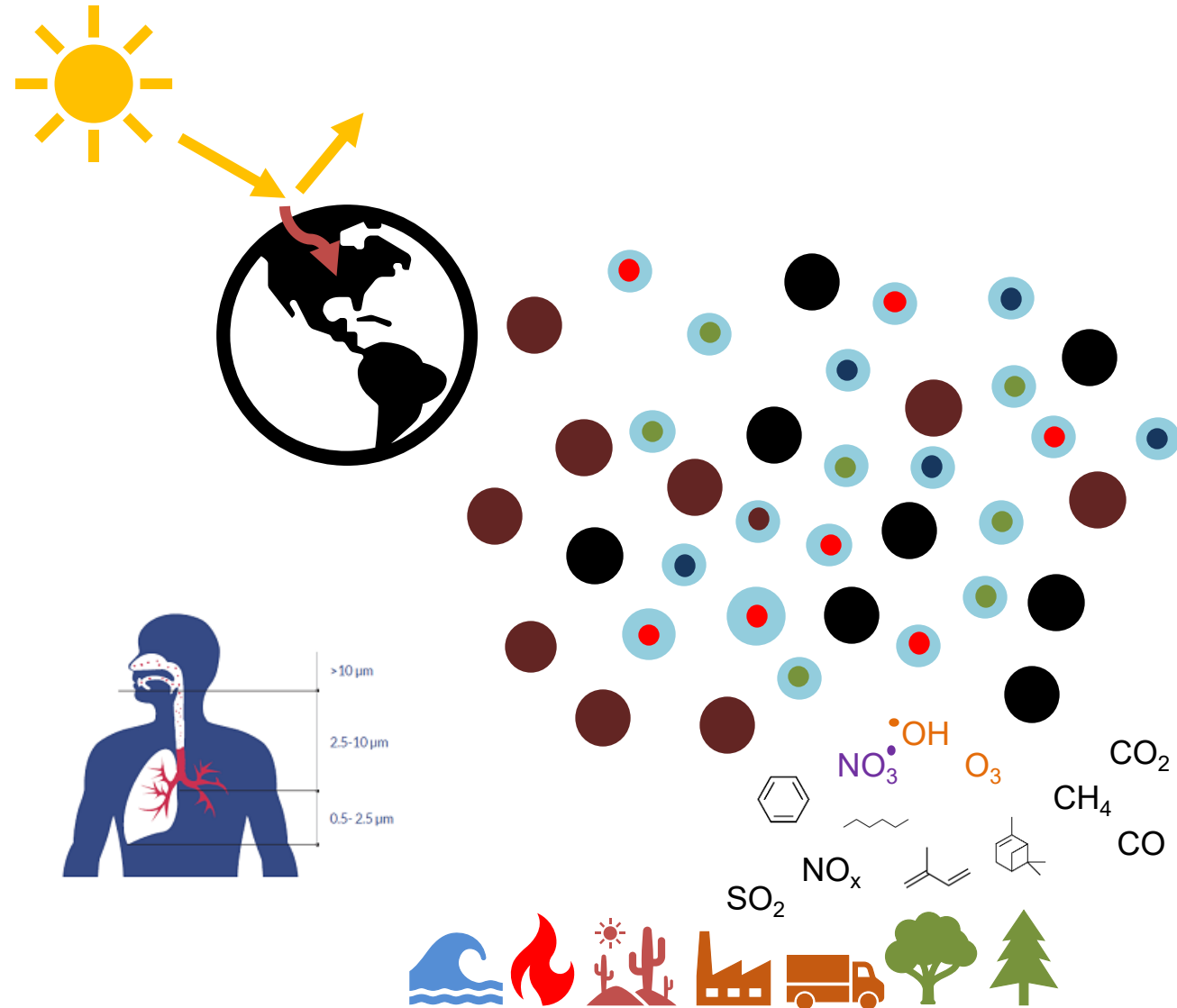
Approximately 120 million people nationwide live in counties with pollution levels above regulations



Legend \*\*

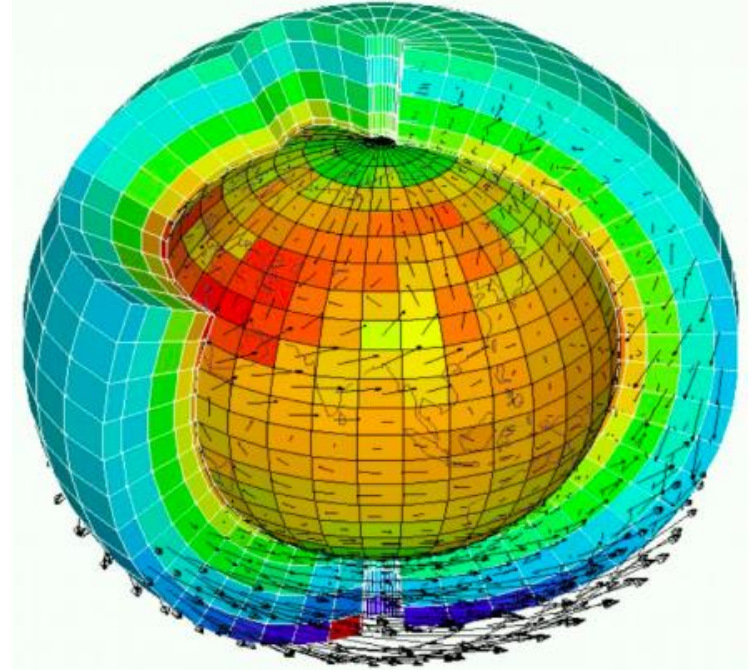
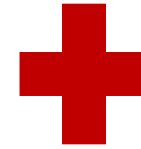
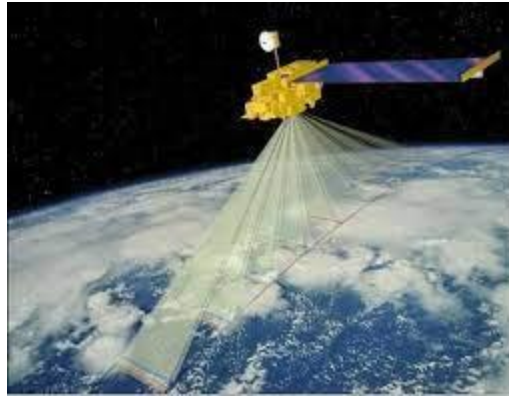
- County Designated Nonattainment for 6 NAAQS Pollutants
- County Designated Nonattainment for 5 NAAQS Pollutants
- County Designated Nonattainment for 4 NAAQS Pollutants
- County Designated Nonattainment for 3 NAAQS Pollutants
- County Designated Nonattainment for 2 NAAQS Pollutants
- County Designated Nonattainment for 1 NAAQS Pollutant

# Fundamental Questions in Atmospheric Chemistry



- What are the spatial distribution and temporal trends of pollutants?
- What drives these trends?
- How can we apply our scientific knowledge of trends and drivers to broader concepts? How do pollutants impact Earth's radiation budget and human health?

# Christiansen Lab General Methods

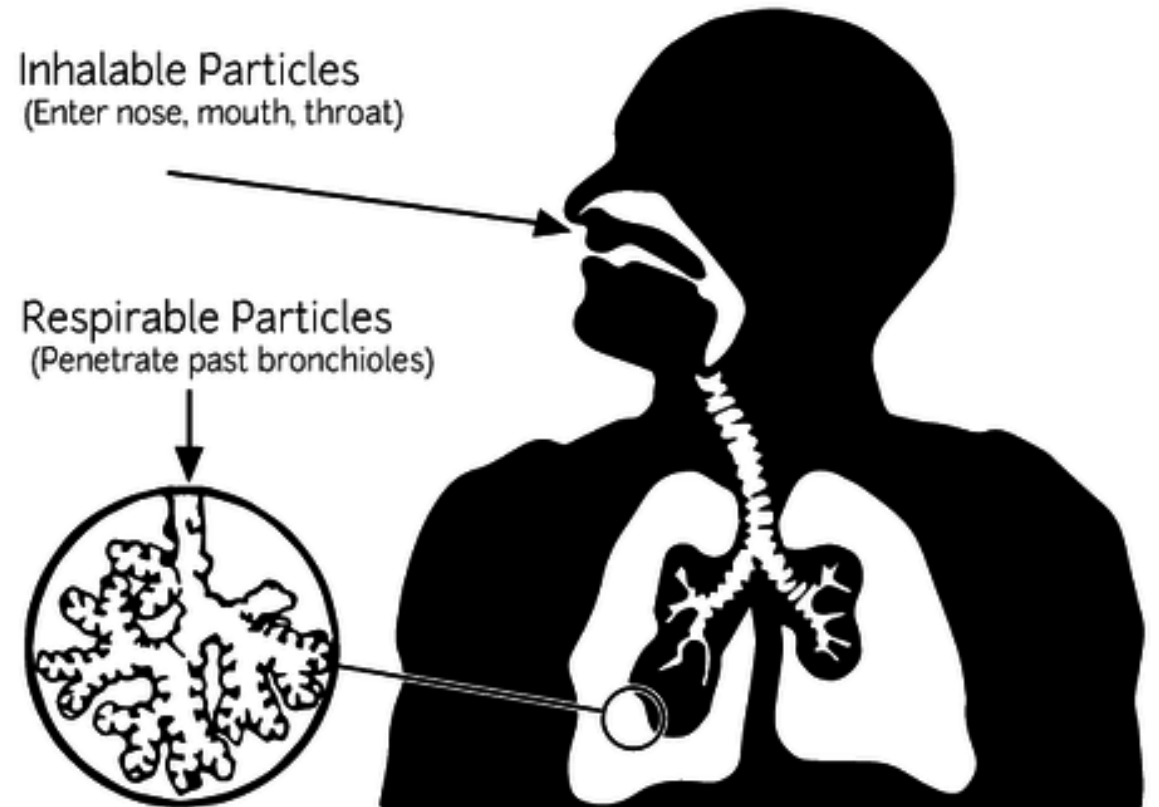


Environmental monitoring datasets (ground measurements, vertical profiles, satellites)  
Atmospheric models

- 1) Combine datasets in unique ways
- 2) Combine measurements with models to help understand interactions
- 3) Use measurements to improve models

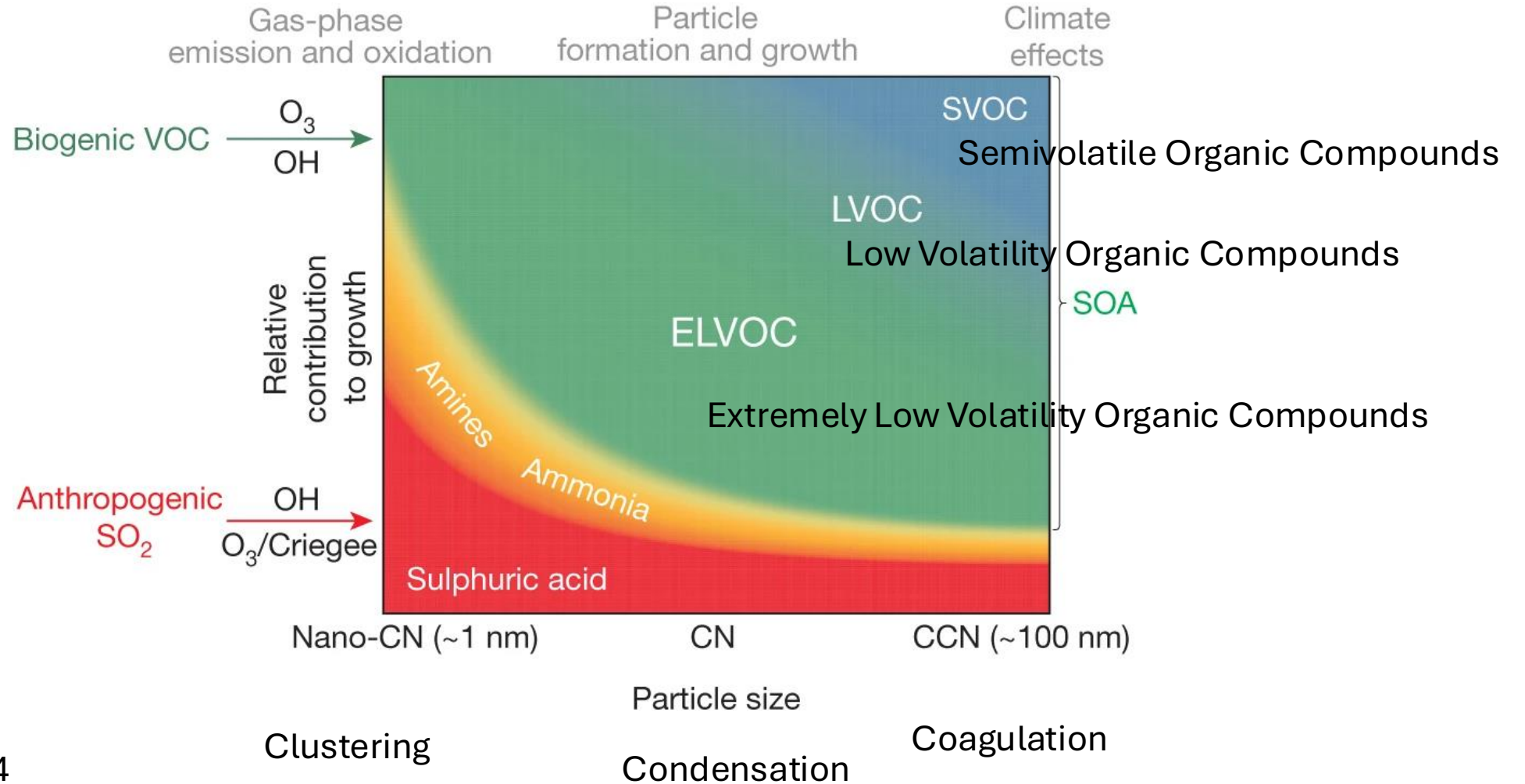
# Why are air pollutants harmful?

- Some particles can get past protective layer of lungs (cilia) and enter bloodstream
  - Oxidative stress, inflammatory response → disease



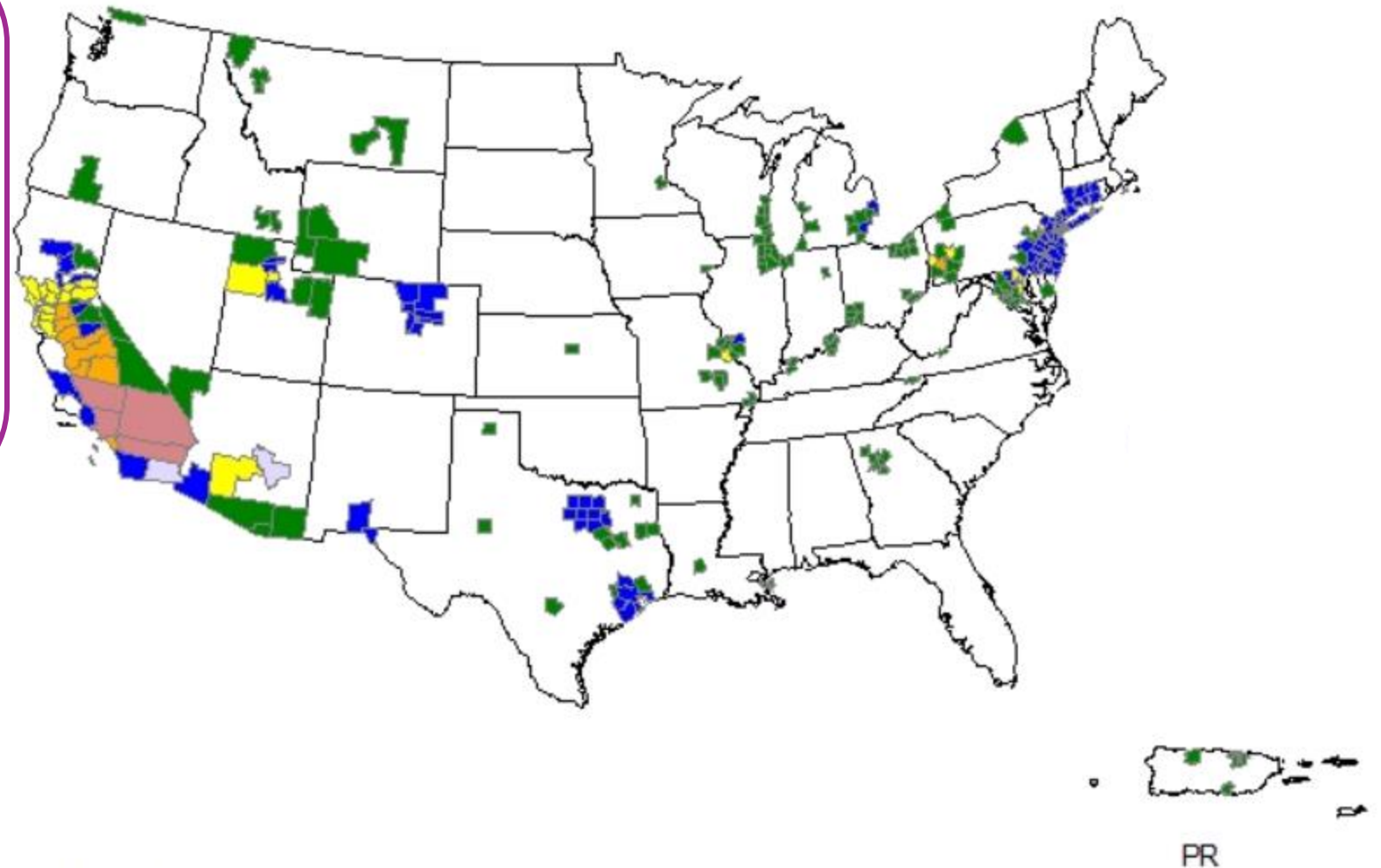
particles in the nose, throat, and lungs  
an edited version of Figure 4-1 EPA/600/R-95/115

# Organics contribute greatly to growth of new particles



Air pollution problems persist today

Approximately 120 million people nationwide live in counties with pollution levels above regulations

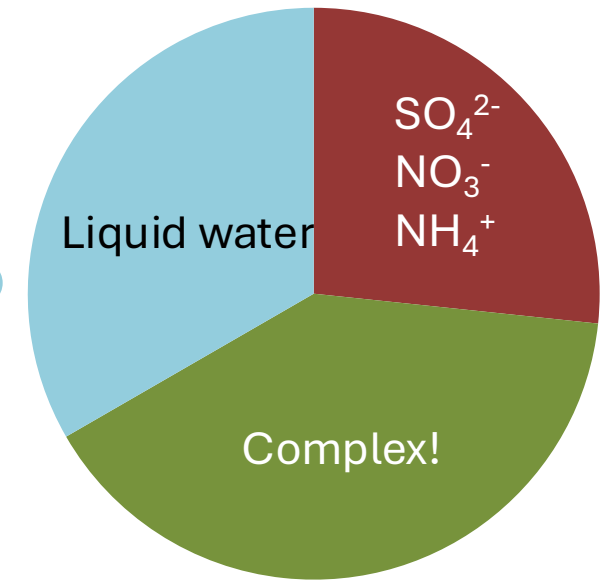
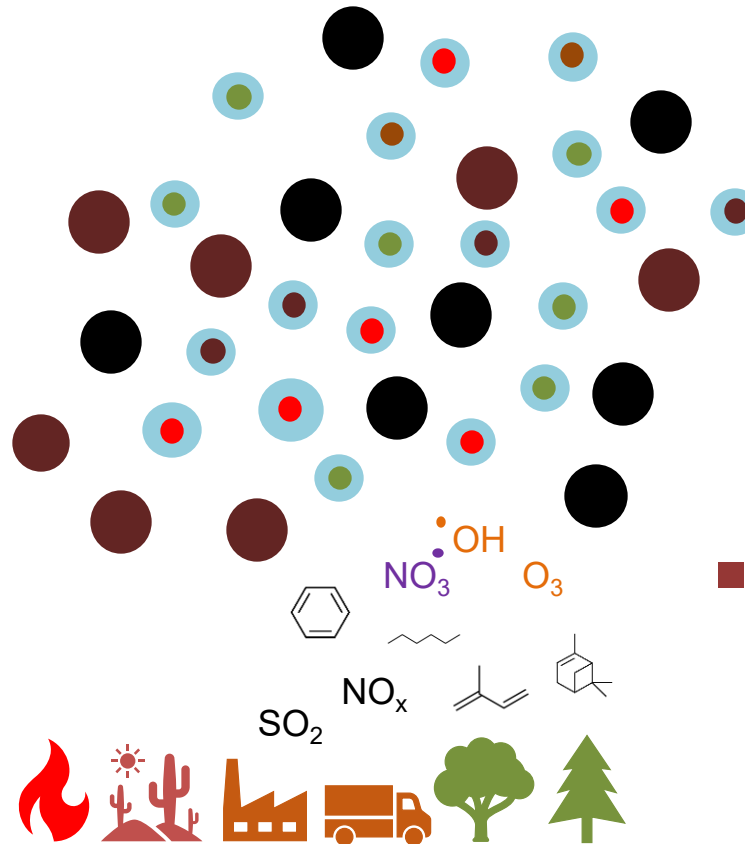


Legend \*\*

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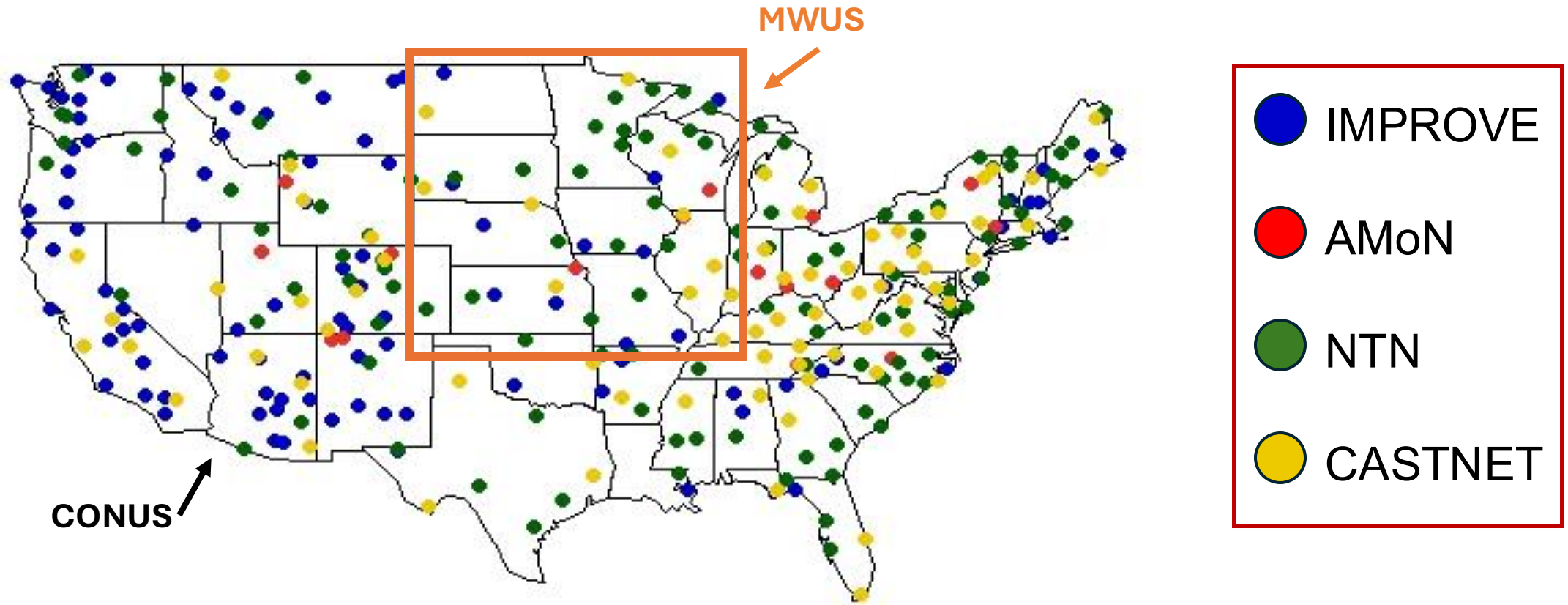
# What is particulate matter (PM)?

- Small solid or liquid particles that are suspended in air
- Primary and secondary sources
  - Primary: Dust, sea spray, soot
  - Secondary: Oxidation of gas-phase precursor species
- Composed of complex chemical matrix



■ Inorganic ■ Organic ■ Water

# Observational Data (2007-2019)



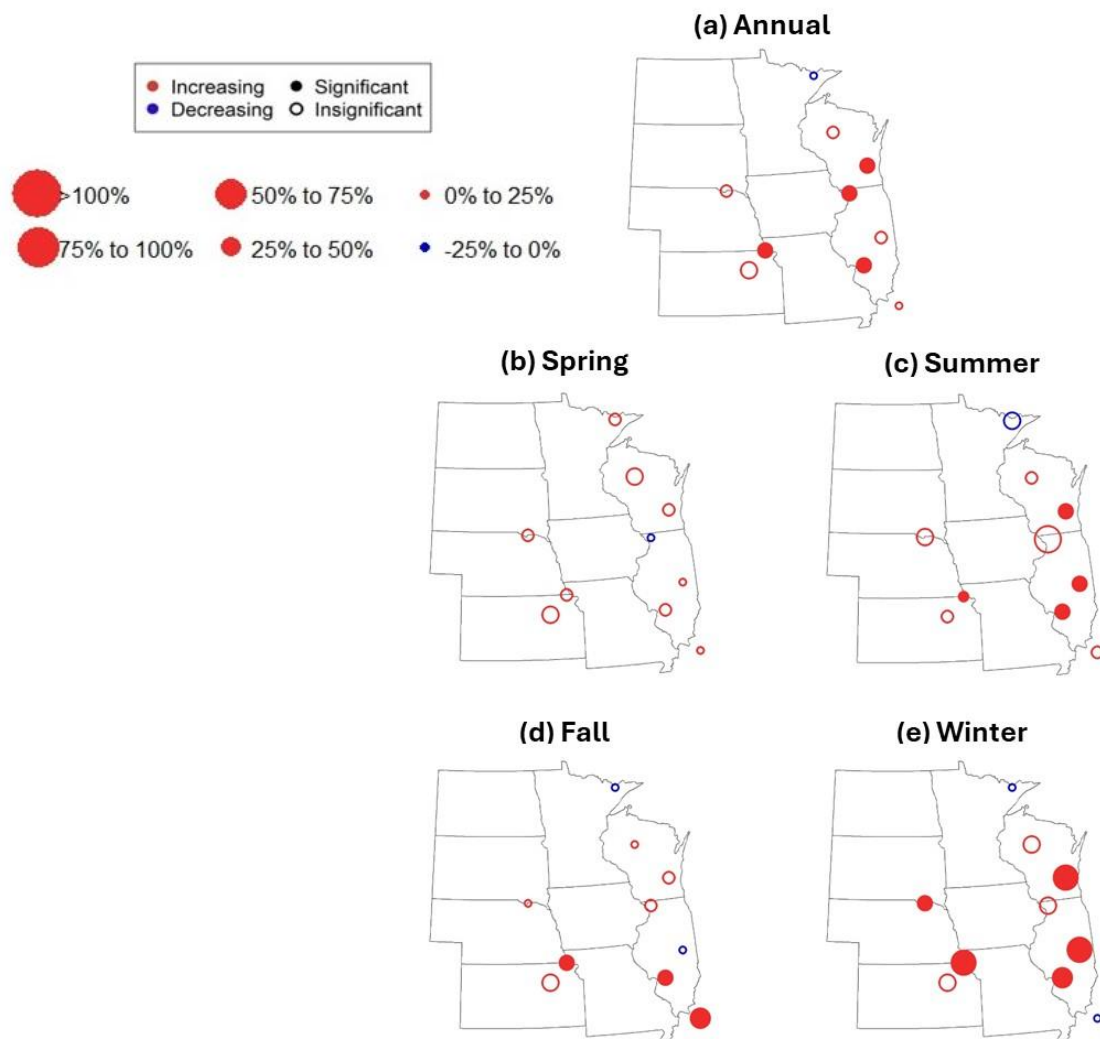
**IMPROVE** (Interagency Monitoring of PROtected Visual Environments):  $PM_{2.5}$  concentrations and chemical speciation

**AMoN** (Ammonia Monitoring Network):  $NH_3$  concentrations

**NTN and CASTNET** (National Trends Network and Clean Air Status and Trends Network):  $NH_4^+$  and total N deposition

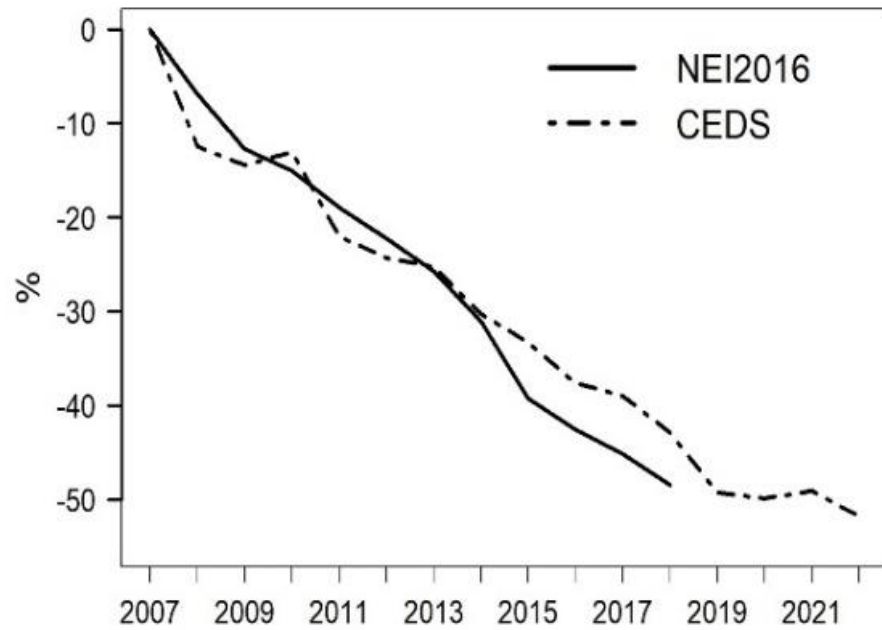
# NH<sub>3</sub> Concentrations Increase in the MWUS Faster than the CONUS

## NH<sub>3</sub> Concentration Trends over the MWUS

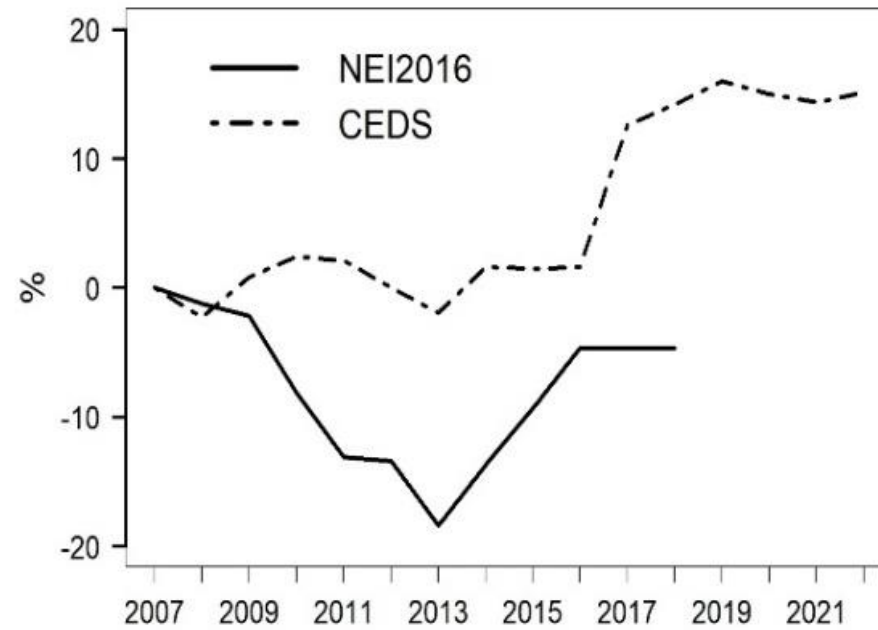


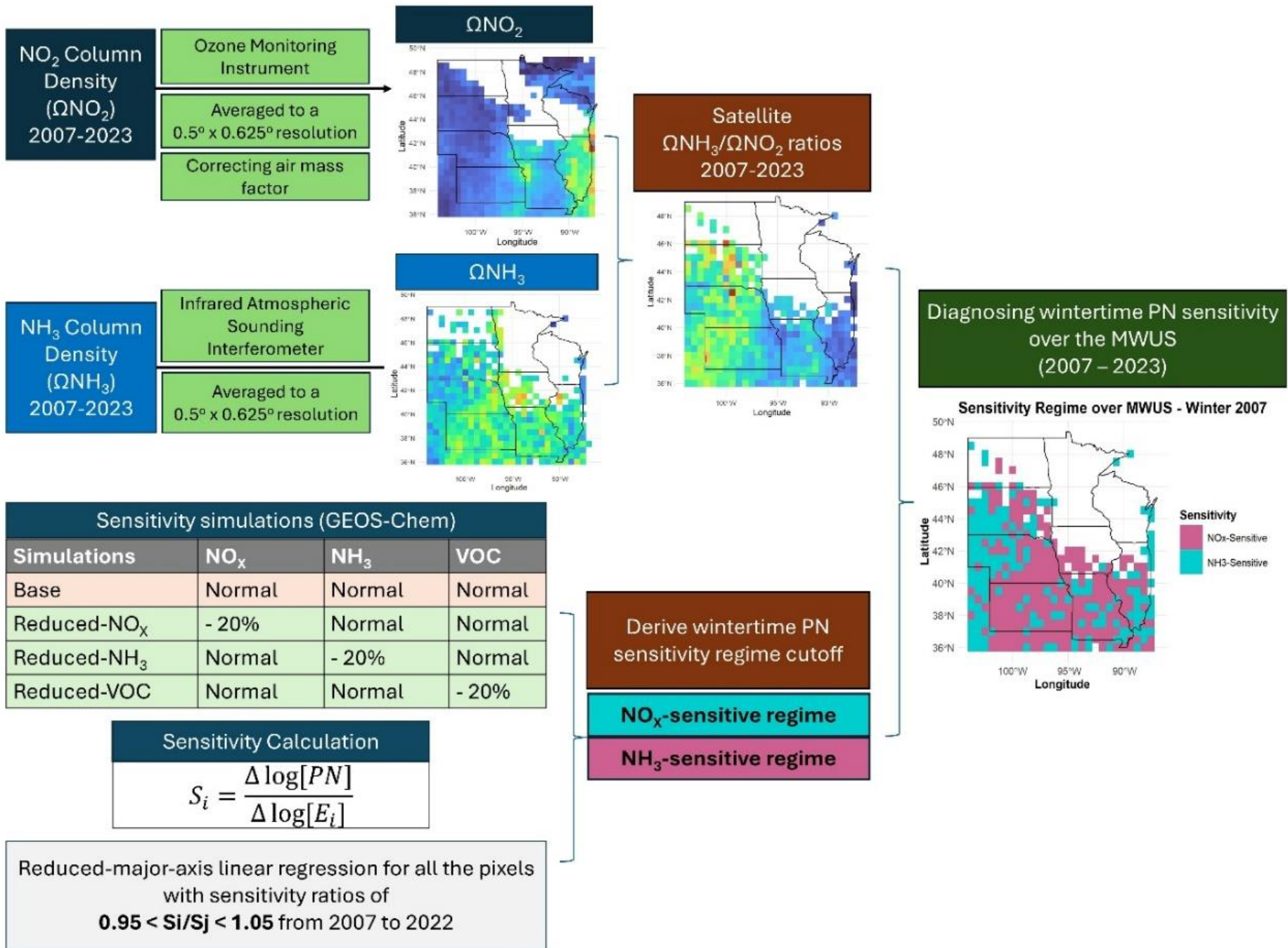
- NH<sub>3</sub> increases by 42% in the MWUS on average from 2007-2019 and are larger than the CONUS average (35%)
  - Winter shows largest increases (63% on average)
- Primarily agriculture: ~95% of total NH<sub>3</sub> emissions in MWUS are from agriculture
  - No correlation with meteorological variables
  - Perhaps due to nitrogen fertilizer overuse or unreacted NH<sub>3</sub>

MWUS Wintertime : NO<sub>x</sub> Emissions (2007 – 2022) (NEI2016 vs. CEDS)



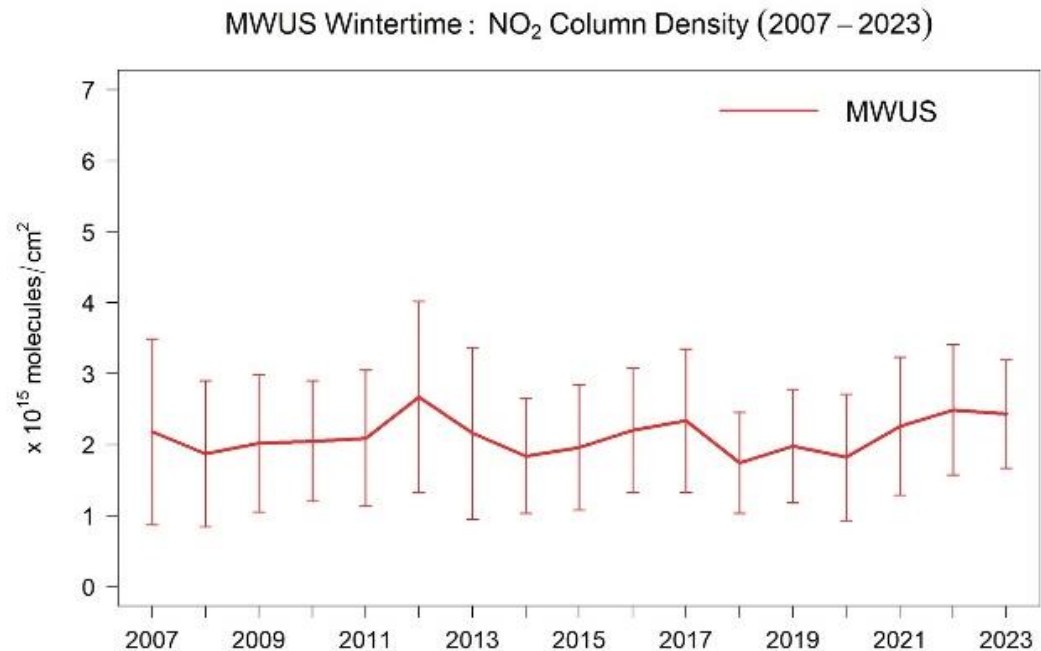
MWUS Wintertime : NH<sub>3</sub> Emissions (2007 – 2022) (NEI2016 vs. CEDS)





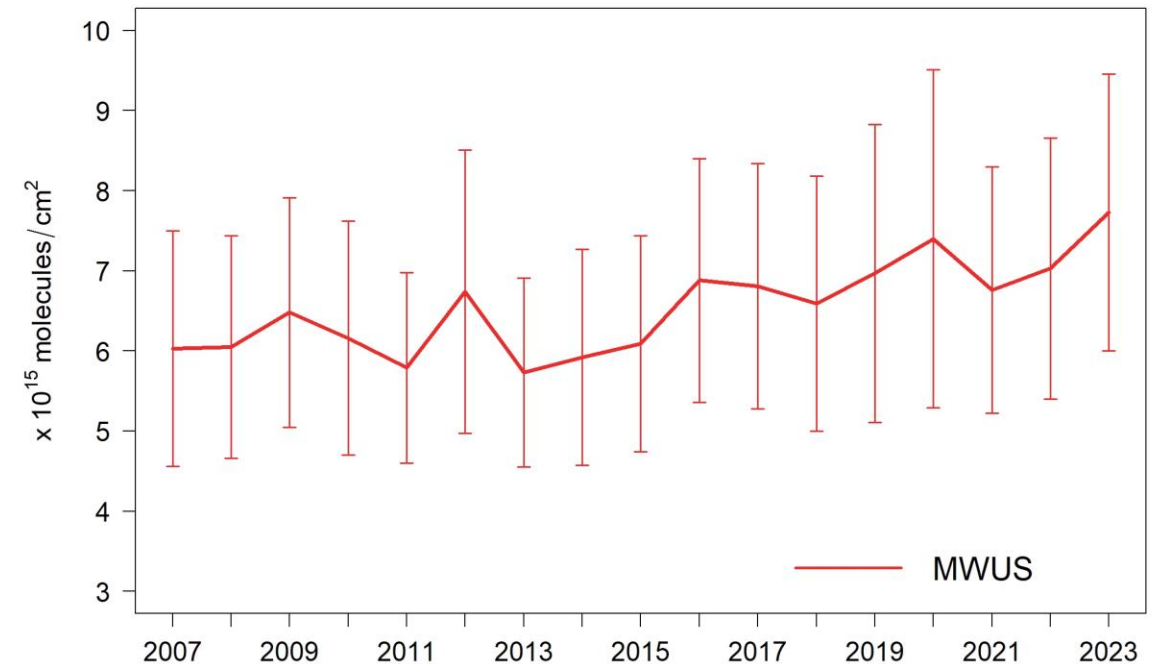
# NO<sub>2</sub> and NH<sub>3</sub> Column Density Trends over MWUS (2007–2023)

## NO<sub>2</sub> Column Densities (2007-2023)



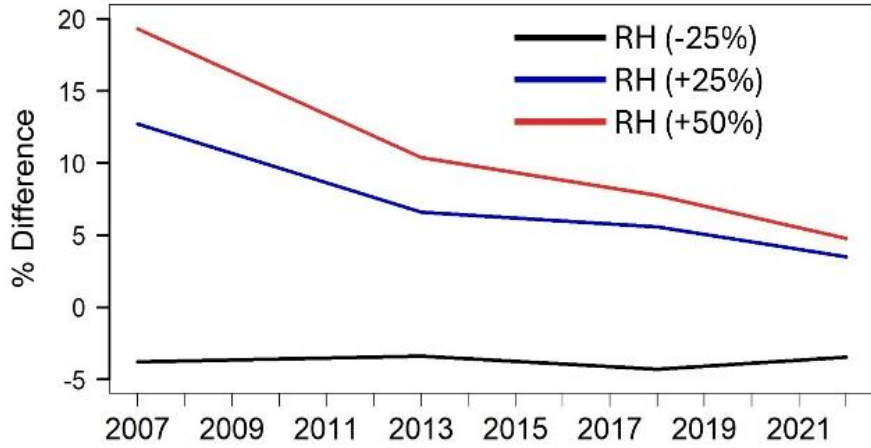
The flat trend in regional NO<sub>2</sub> is driven by natural sources and free tropospheric NO<sub>2</sub>

## NH<sub>3</sub> Column Densities (2007-2023)

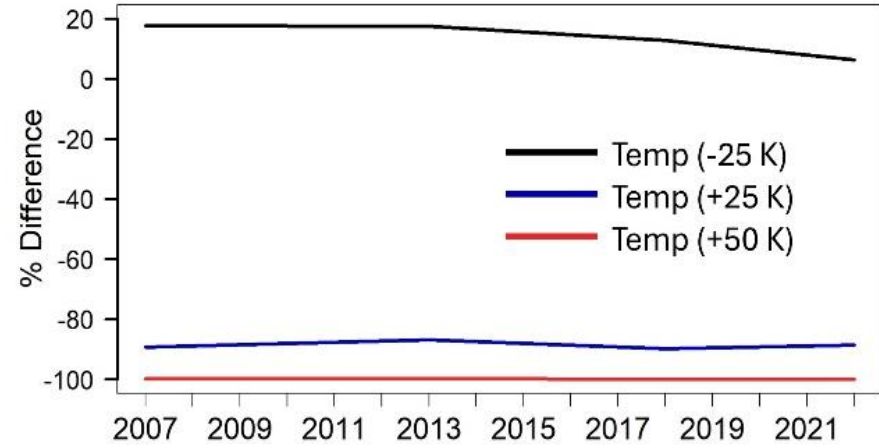


The increase in NH<sub>3</sub> column densities is partly attributable to increases in agricultural activities

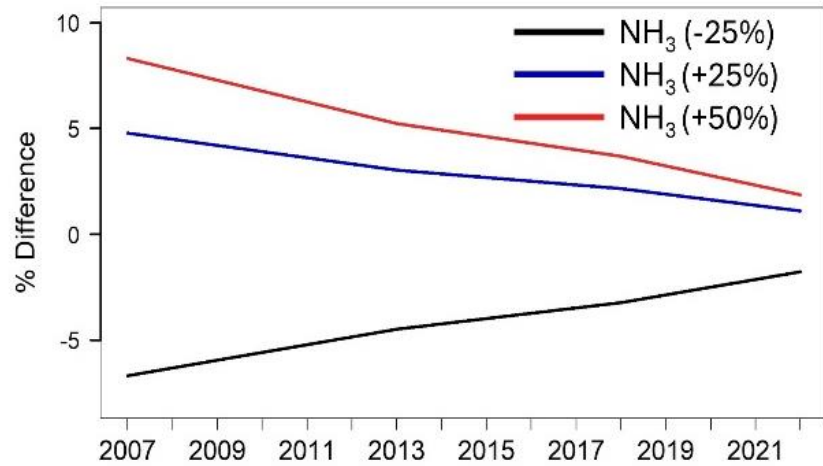
MWUS Winter : Changes in PN vs. Changes in RH (ISORRPIA)



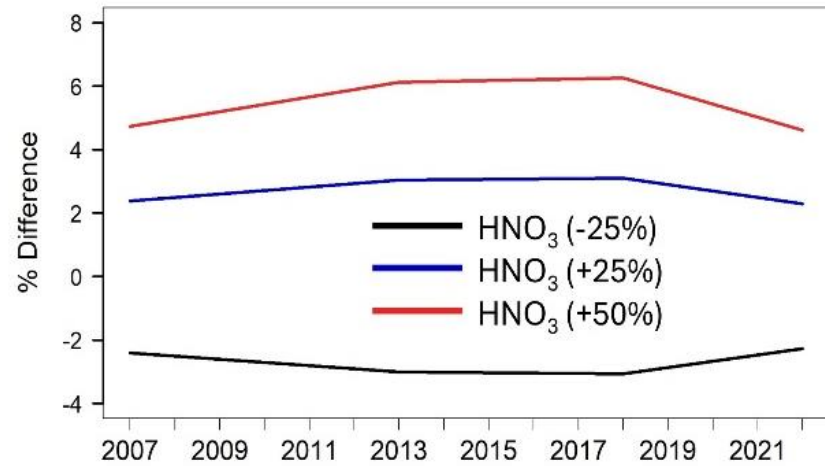
MWUS Winter : Changes in PN vs. Changes in Temp (K) (ISORRPIA)



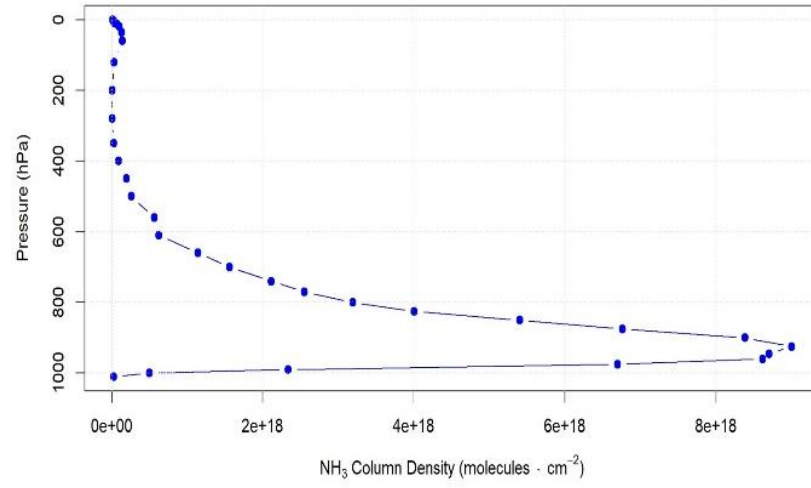
MWUS Winter : Changes in PN vs. Changes in NH<sub>3</sub> Concentrations (ISORRPIA)



MWUS Winter : Changes in PN vs. Changes in HNO<sub>3</sub> Concentrations (ISORRPIA)



NH<sub>3</sub> Vertical Profile



NO<sub>2</sub> Vertical Profile

